

Summer school on Epitaxy

MATEPI 2025

Porquerolles June 22-27 2025

Chapter II Epitaxy : experimental description

II.6 Comparison of oxide and semiconductor epitaxy and oxide/semiconductor integration

Guillaume Saint-Girons

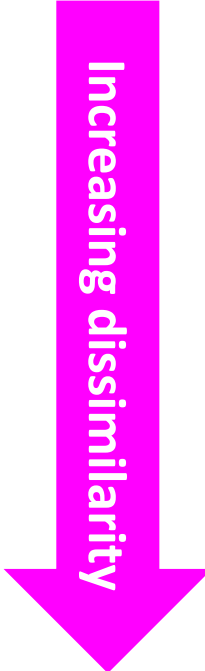
INL-UMR5270/CNRS, Ecole Centrale de Lyon, 36 avenue Guy de Collongue 69134 Ecully cedex, France



Outlook

Introduction

- a. Accommodation/growth mode and material-substrate dissimilarity
- b. Oxides and semiconductors : comparison of some properties relevant to epitaxy



Increasing dissimilarity

Strained 2D growth and plastic relaxation

- a. Dislocation mediated plastic relaxation process
- b. Plastic relaxation : oxides vs semiconductors
- c. Some exotic effects of oxide epitaxy related to epitaxial strain

Strain free 3D growth

- a. Strain-induced 3D growth: Stranski-Krastanov growth mode
- b. Interface induced 3D growth

Highly dissimilar epitaxial systems

- a. Indirect epitaxial relationships
- b. Mismatch accommodation via interfacial dislocation networks
- c. Interface chemical reactions

Conclusion and future challenges

- a. Oxide/Si templates: a mature technology for integration
- b. MBE for oxide growth

Introduction

Accommodation/growth mode and material-substrate dissimilarity

Condensation of a vapor of A on a substrate B : energy balance

$\Delta\mu$: Chemical potential difference

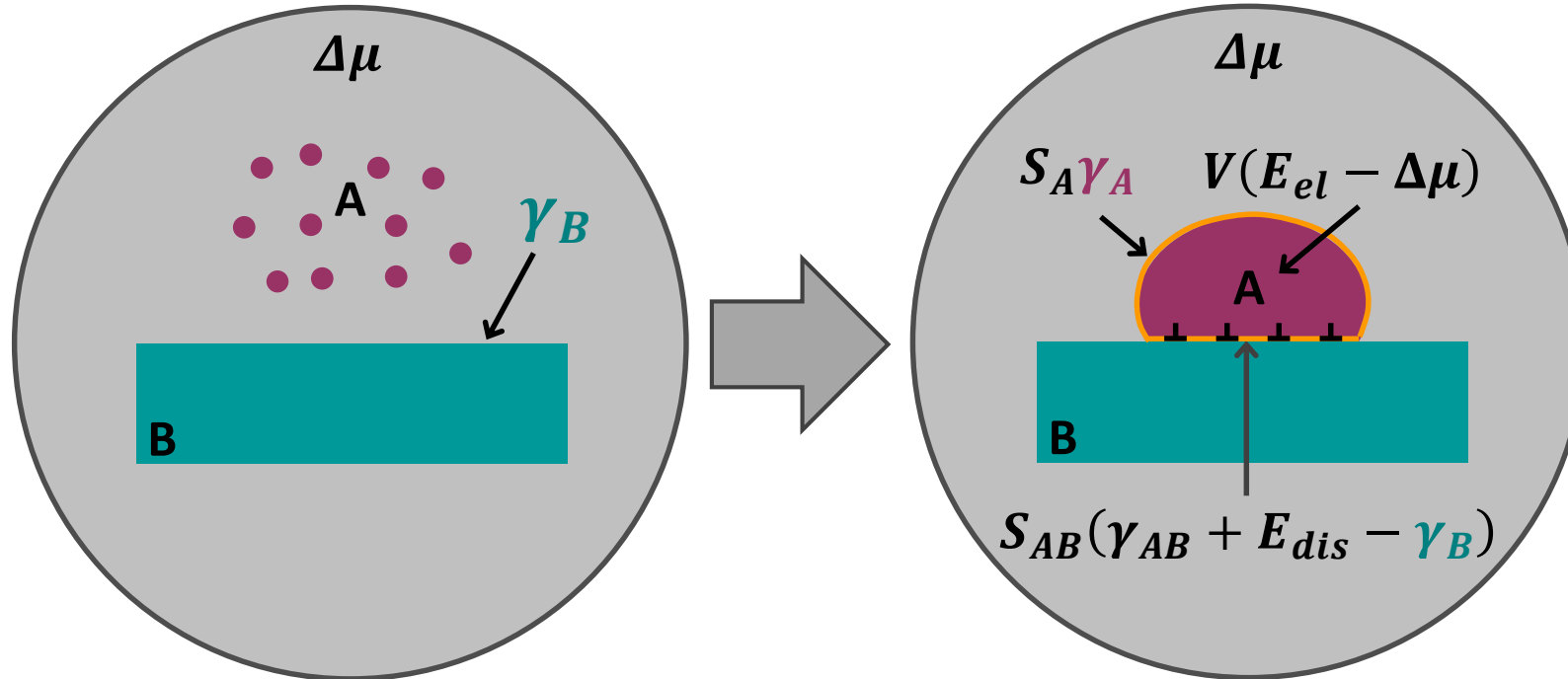
γ_A : Material surface energy

γ_B : Substrate surface energy

γ_{AB} : Interface energy

E_{dis} : Interface dislocation energy

E_{el} : Elastic energy



V : Droplet volume

S_A : Droplet surface

S_{AB} : Interface area

$$\text{Nucleation free enthalpy : } \Delta G = V(E_{el} - \Delta\mu) + S_{AB}(\gamma_{AB} + E_{dis} - \gamma_B) + S_A\gamma_A$$

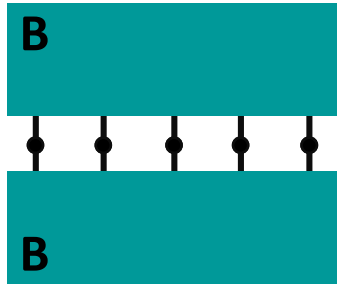
Introduction

Accommodation/growth mode and material-substrate dissimilarity

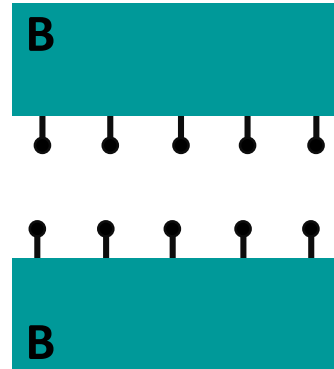
Surface and interface energies

I. Markov, Mat. Chem. Phys. **36**, 1 (1993)

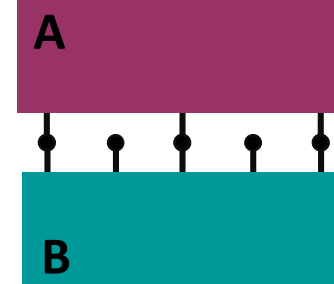
N_B B-B bonds/ m^2
Energy E_{BB}



$N_B E_{BB}$



$-\beta$



Energy cost E_{tot}

$$N_B E_{BB} + N_A E_{AA} - 2x N_B E_{AB}$$

Interface energy γ_{AB}

$$\gamma_{AB} = \frac{E_{tot}}{2} = \gamma_A + \gamma_B - \beta$$

$$= N_B \left[\frac{x E_{AA}}{2} + \frac{E_{BB}}{2} - \min(x, 1) E_{AB} \right]$$

Notwithstanding
surface relaxation
mechanisms $\longrightarrow$

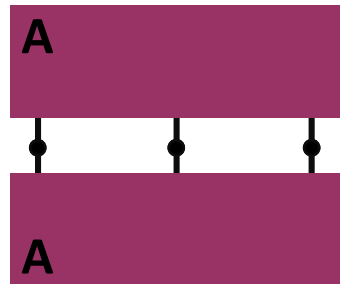
Surface energy γ_B (γ_A)

$$\gamma_B \sim \frac{N_B E_{BB}}{2}$$

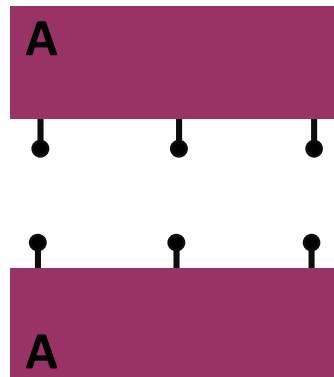
Adhesion energy β

$$x N_B E_{AB} \text{ if } x < 1$$

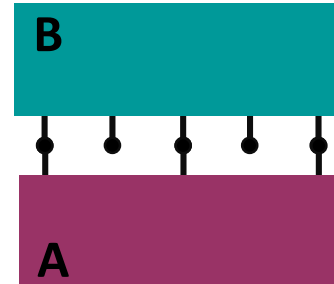
$$N_B E_{AB} \text{ if } x > 1$$



$N_A E_{AA}$



$-\beta$



Homoepitaxy: $\gamma_{AB} = 0$

Strong adhesion: $\gamma_{AB} < 0$

$$\frac{x E_{AA} + E_{BB}}{2} < \min(x, 1) E_{AB}$$

Low adhesion: $\gamma_{AB} > 0$

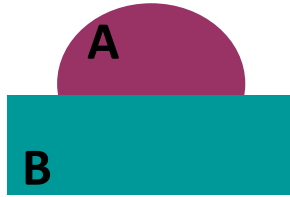
$$\frac{x E_{AA} + E_{BB}}{2} > \min(x, 1) E_{AB}$$

$N_A = x N_B$ A-A bonds/ m^2
Energy E_{AA}

Introduction

Accommodation/growth mode and material-substrate dissimilarity

Elastic energy



$$E_{el} = \frac{Y_A}{1 - \nu_A} m^2 \left(1 - e^{-2k/p} \right)$$

Y_A : Young modulus

ν_A : Poisson ratio

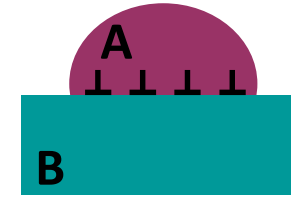
$m = \frac{a_A - a_B}{a_B}$: lattice mismatch

p : droplet aspect ratio

k : 0.073 for a flat cylinder, 0.082 for a spherical cap

K. Tillmann and A. Förster, Thin Solid Films 368, 93, (2000)

Interfacial dislocation network energies



$$E_{dis} = mb \frac{Y_A Y_B}{Y_A(1 + \nu_A) + Y_B(1 + \nu_B)} \left(\frac{1}{4\pi} \ln \frac{R_c}{b} + 0.1 \right)$$

b : Burger vector norm

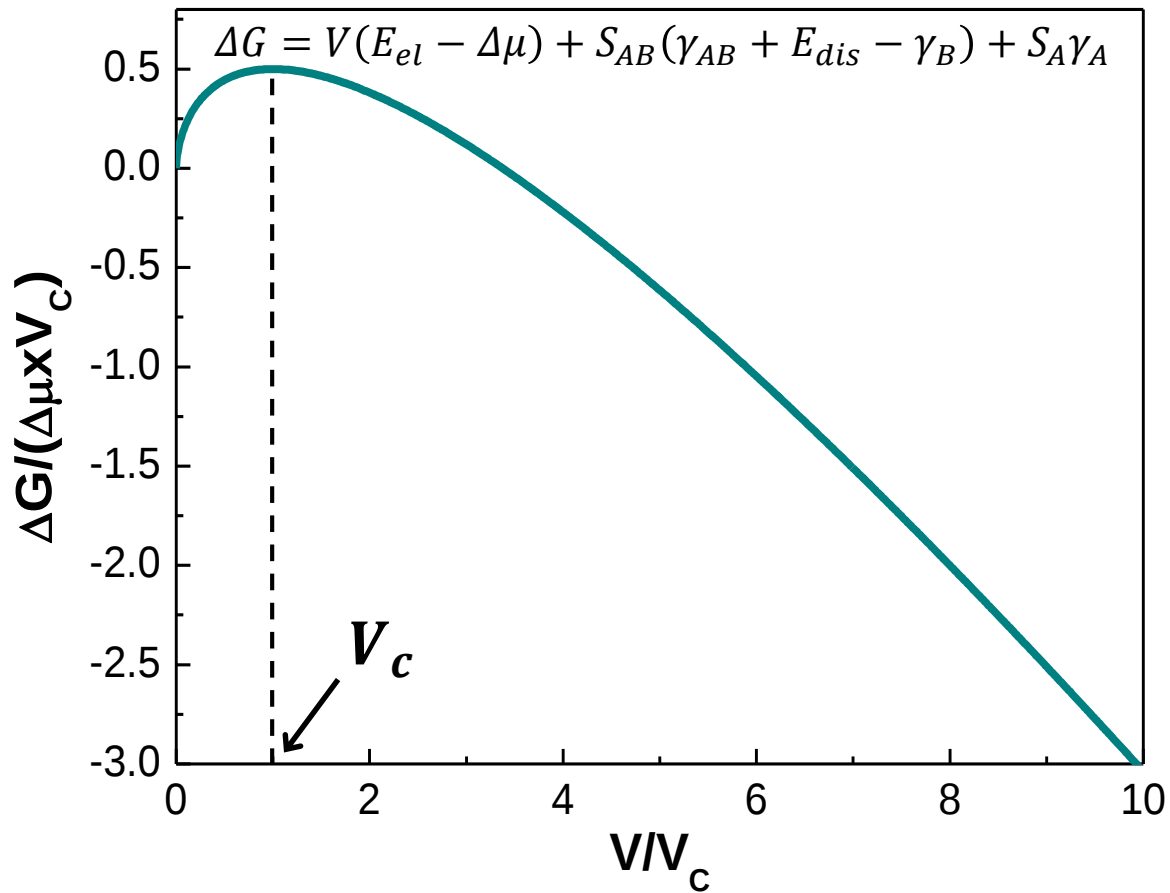
R_c : Cut-off radius (1/2 distance between 2 successive dislocations or droplet height)

Introduction

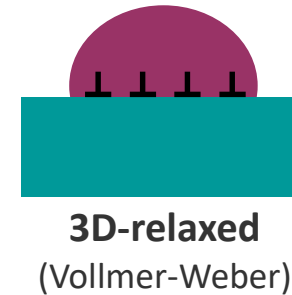
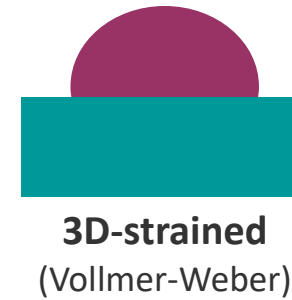
Accommodation/growth mode and material-substrate dissimilarity

Critical volume V_c

No stable nucleus with $V < V_c$

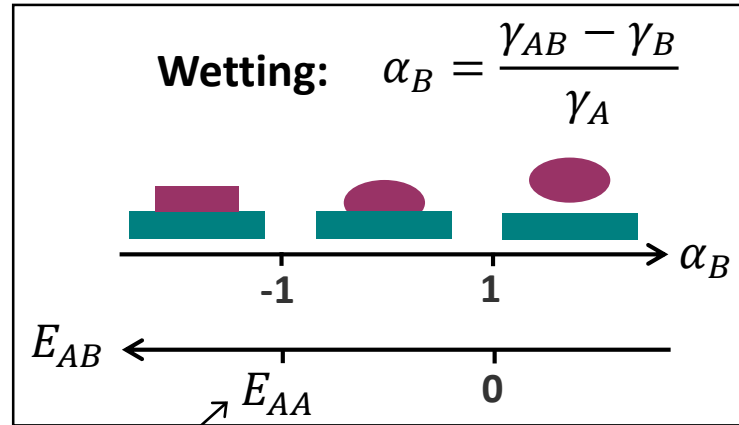


The observed configuration is that minimizing V_c



Introduction

Accommodation/growth mode and material-substrate dissimilarity



$x E_{AA}$
if $x > 1$

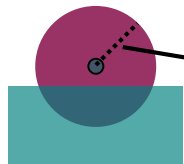
Critical volume V_c

Geometrical factors

$$V_c = \left(\frac{2}{3} L \frac{\alpha_B f + g}{1 - \alpha_C} \right)^3$$

Nucleus radius (@ 0 strain):

$$L = \frac{\gamma_A}{\Delta\mu}$$



if $\alpha_C = 0$

Strain: $\alpha_C = \frac{E_{el}}{\Delta\mu}$

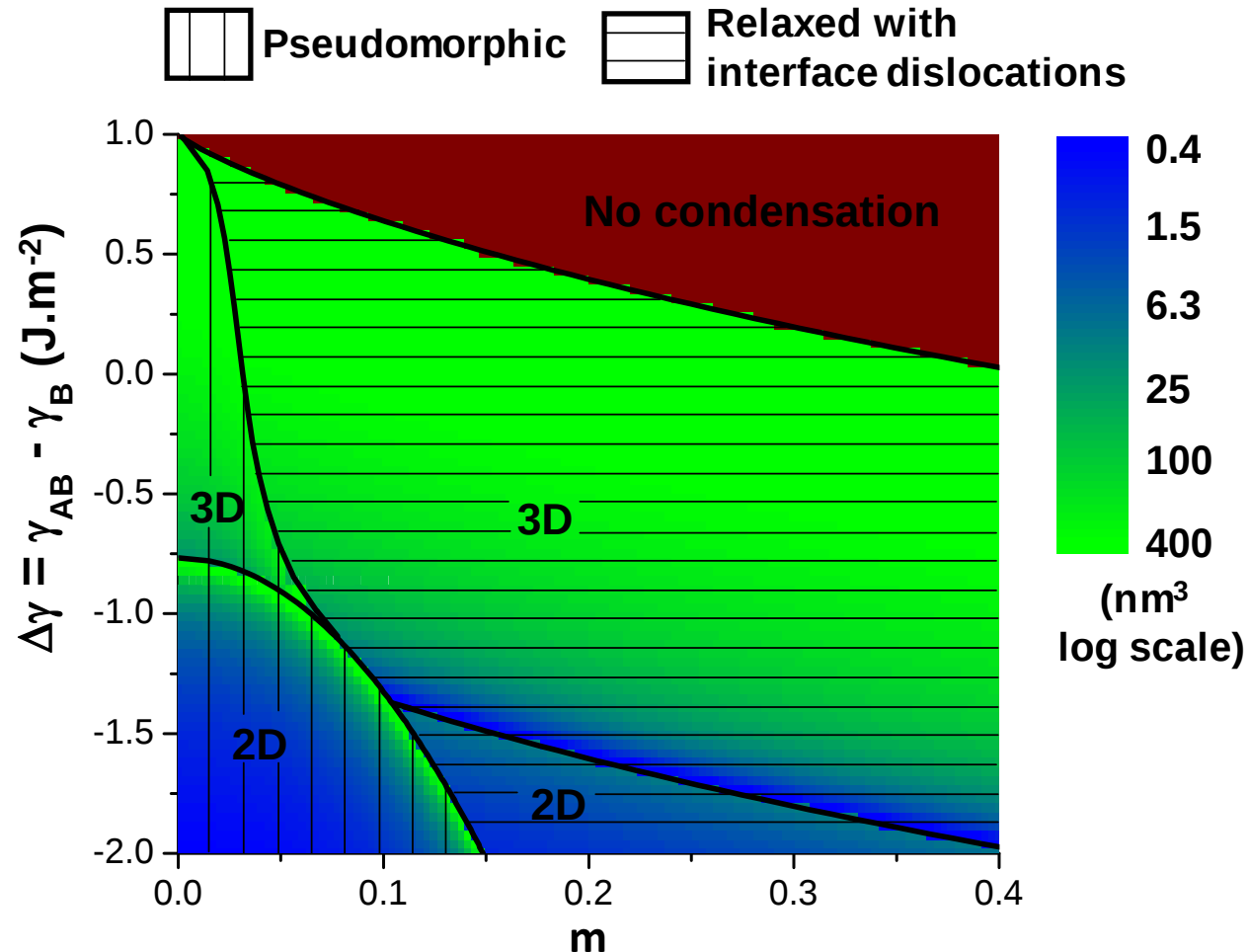
$\alpha_C > 1$: no condensation

Introduction

Accommodation/growth mode and material-substrate dissimilarity

Phase diagram for epitaxy ($\gamma_A = 1 \text{ J/m}^2$)

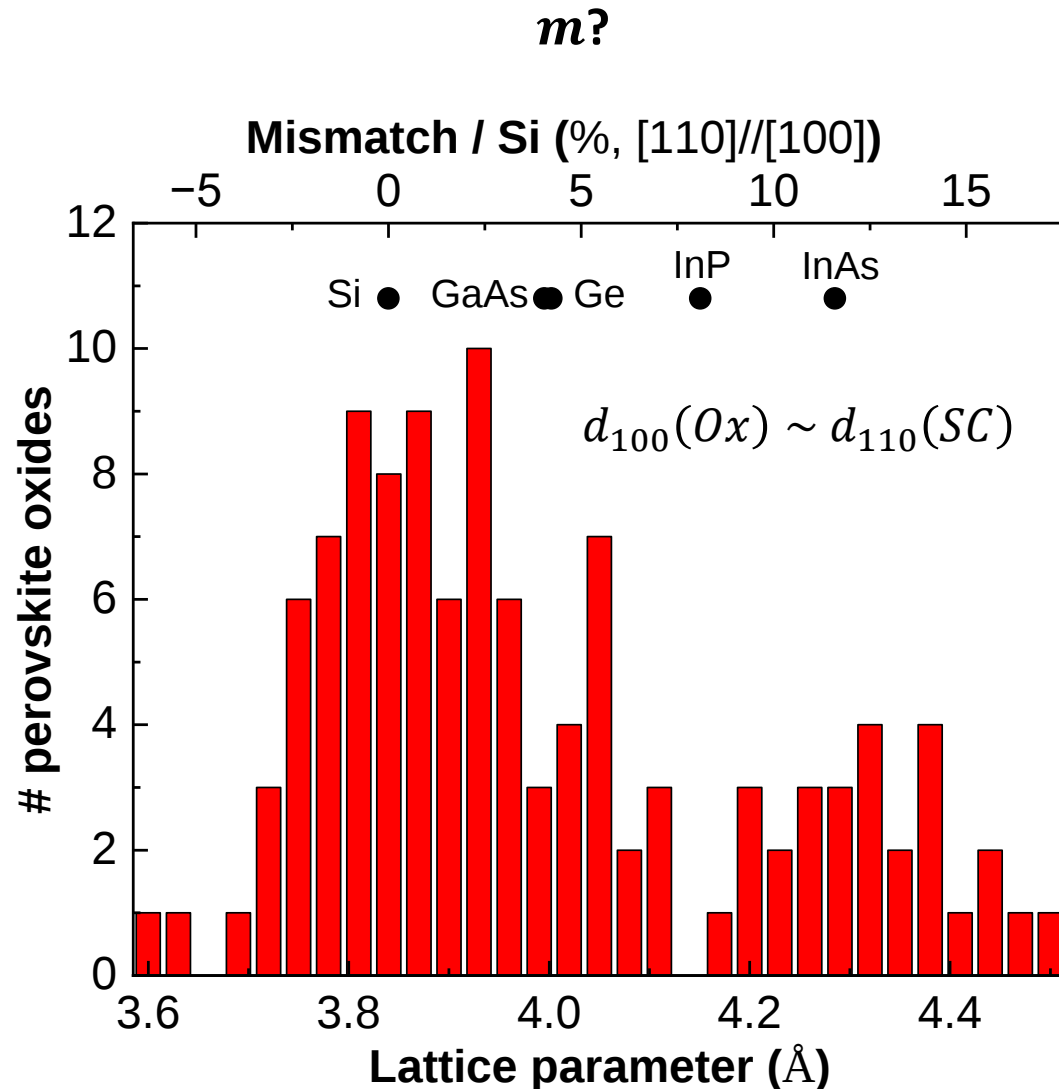
*Saint-Girons et al., Phys. Rev. B **80**, 155308 (2009)*



Where on this diagram are the Ox/Ox, SC/SC and Ox/SC epitaxial systems?

Introduction

Oxide and semiconductors : some properties relevant to epitaxy



$$\gamma_{AB} - \gamma_B?$$

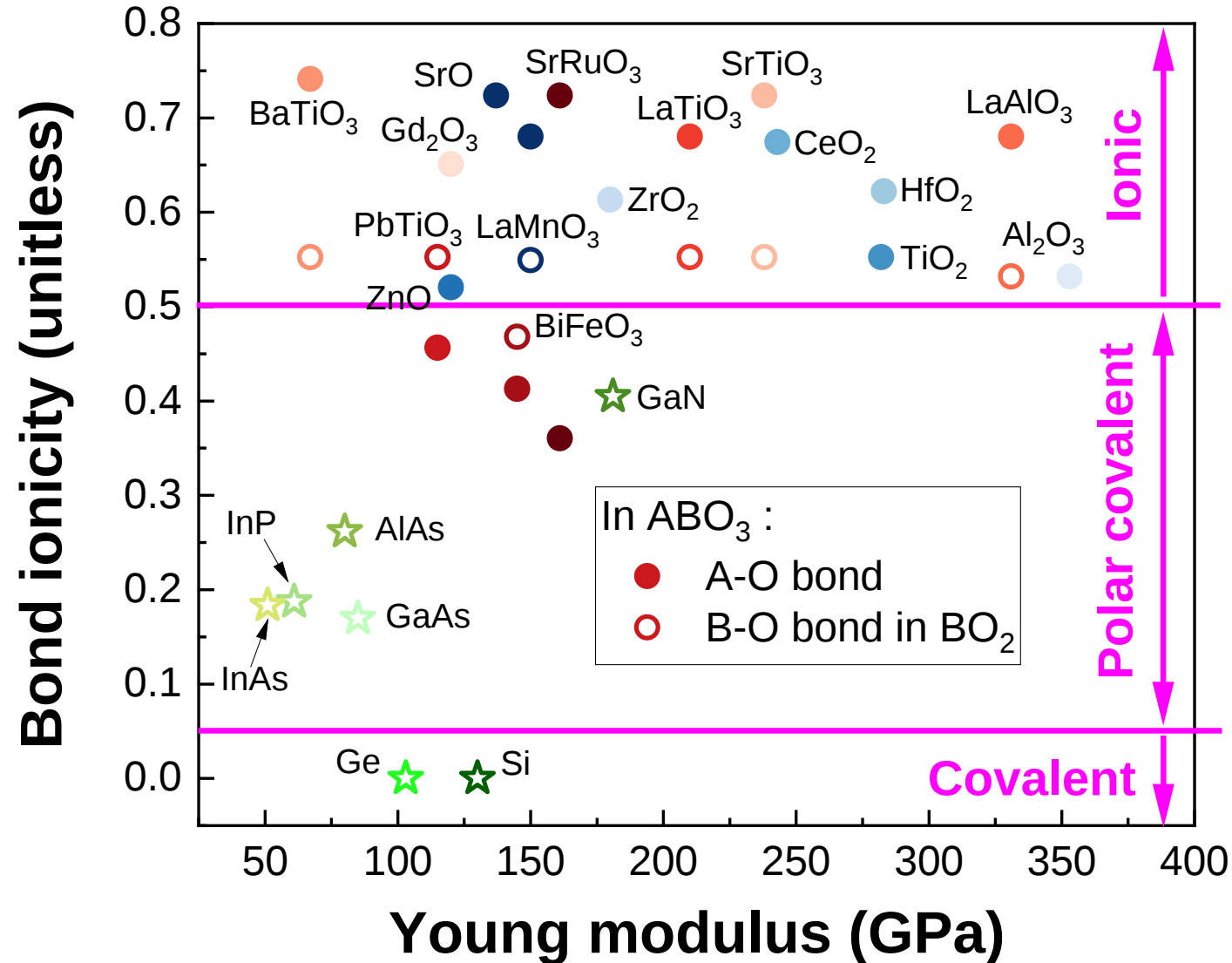
$$\gamma_{AB} - \gamma_B = \gamma_A - \beta \sim N_B \left(\frac{x E_{AA}}{2} - \min(x, 1) E_{AB} \right)$$

$$\gamma_{AB} - \gamma_B \begin{cases} = -\gamma_A \text{ (homoepitaxy)} \\ < -\gamma_A \text{ if } x E_{AB} > \frac{x E_{AA}}{\min(x, 1)} \\ > -\gamma_A \text{ otherwise} \end{cases}$$

γ_A typically ranges from 0.8 to 1.6 J/m²

Introduction

Oxide and semiconductors : some properties relevant to epitaxy



Young modulus:

$Y(\text{Oxides}) > Y(\text{Semiconductors})$

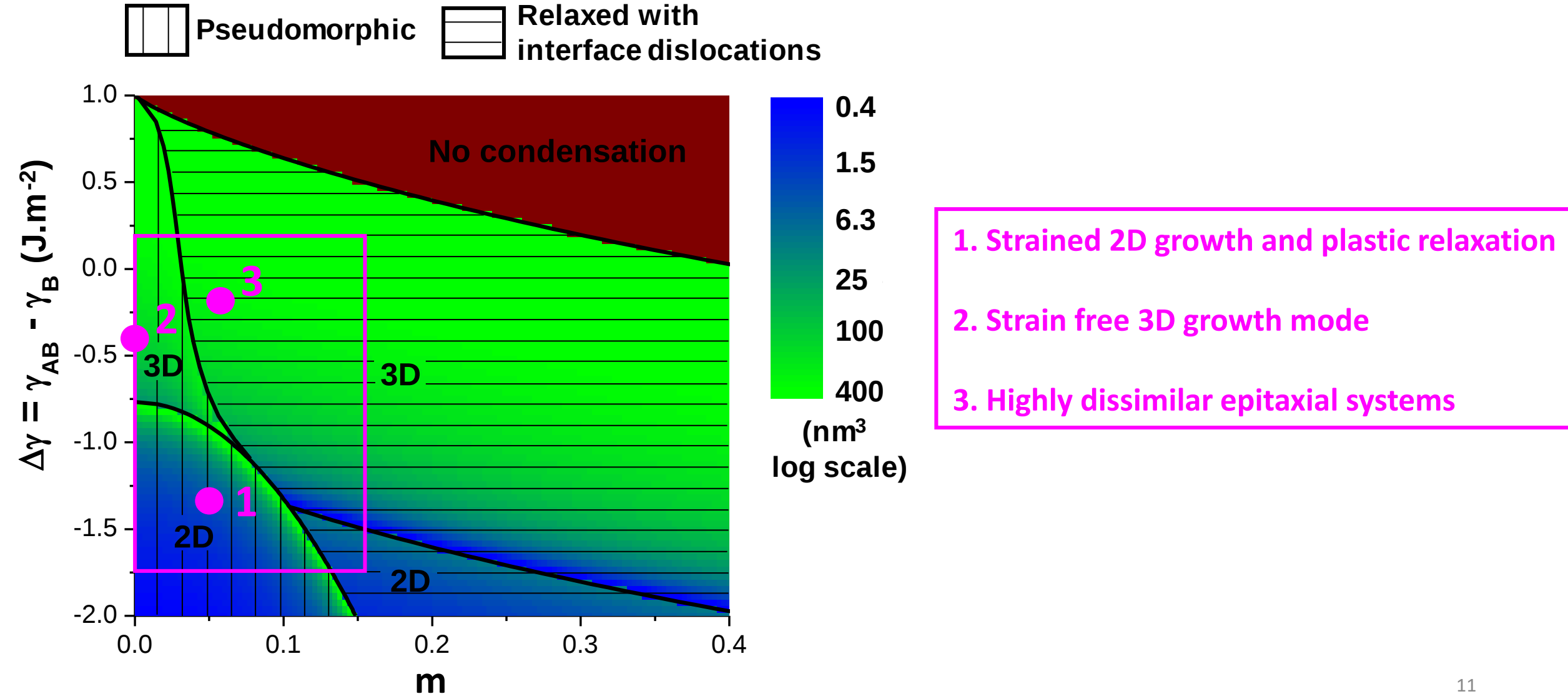
Bond ionicity:

$\Delta EN(\text{Oxides}) > \Delta EN(\text{Semiconductors})$

**Dislocation formation and
propagation hindered in oxides**

Introduction

Exploring the phase diagram with semiconductor and oxide examples

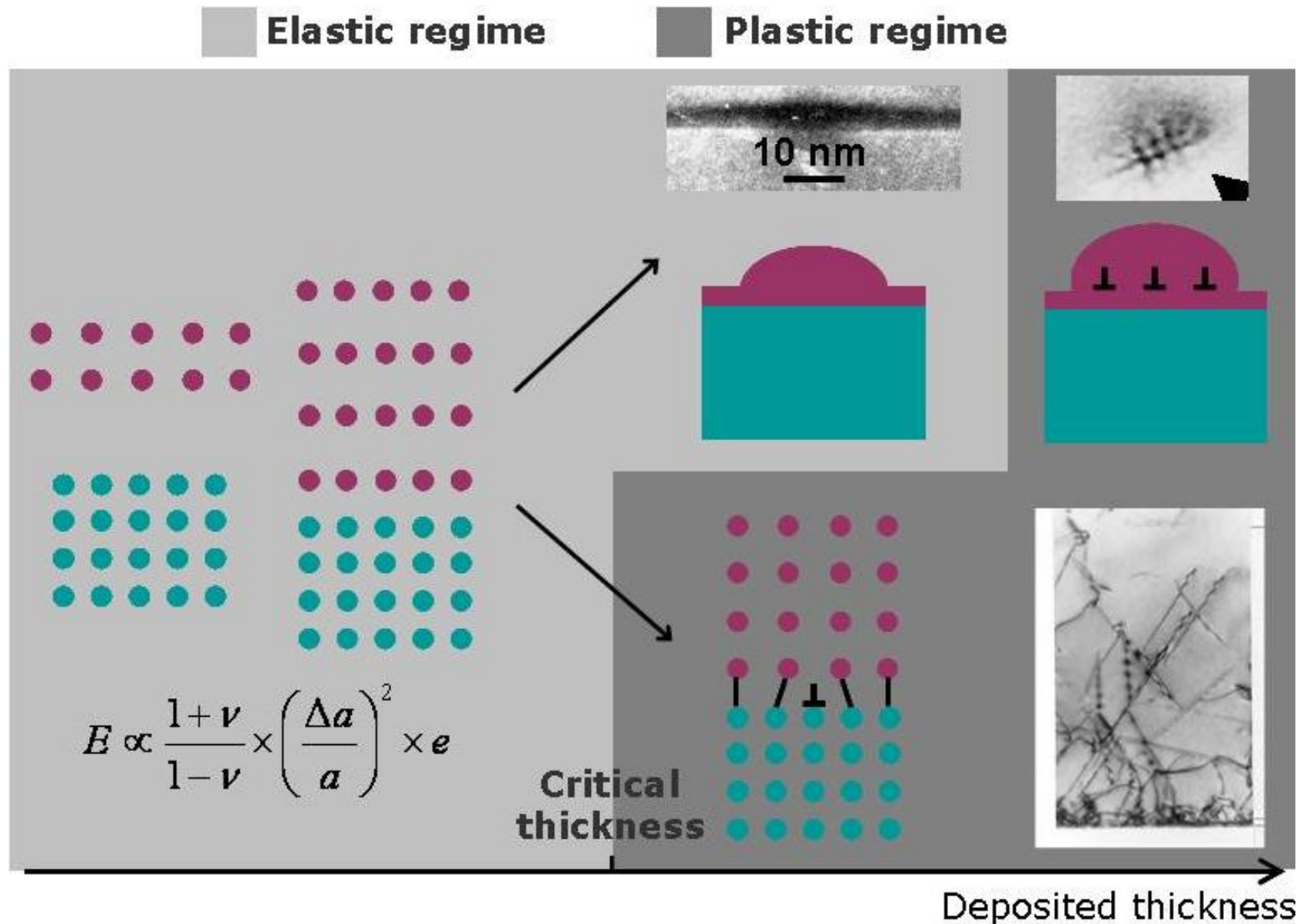


1. Strained 2D growth and plastic relaxation

- a. Dislocation mediated plastic relaxation**
- b. Plastic relaxation: comparison of oxides and semiconductors**
- c. Other effects of epitaxial strain on oxide growth**
 - BaTiO₃ ferroelectric domain structure
 - Strain-assisted phase selection during Carpy-Gally compound growth

Strained 2D growth and plastic relaxation

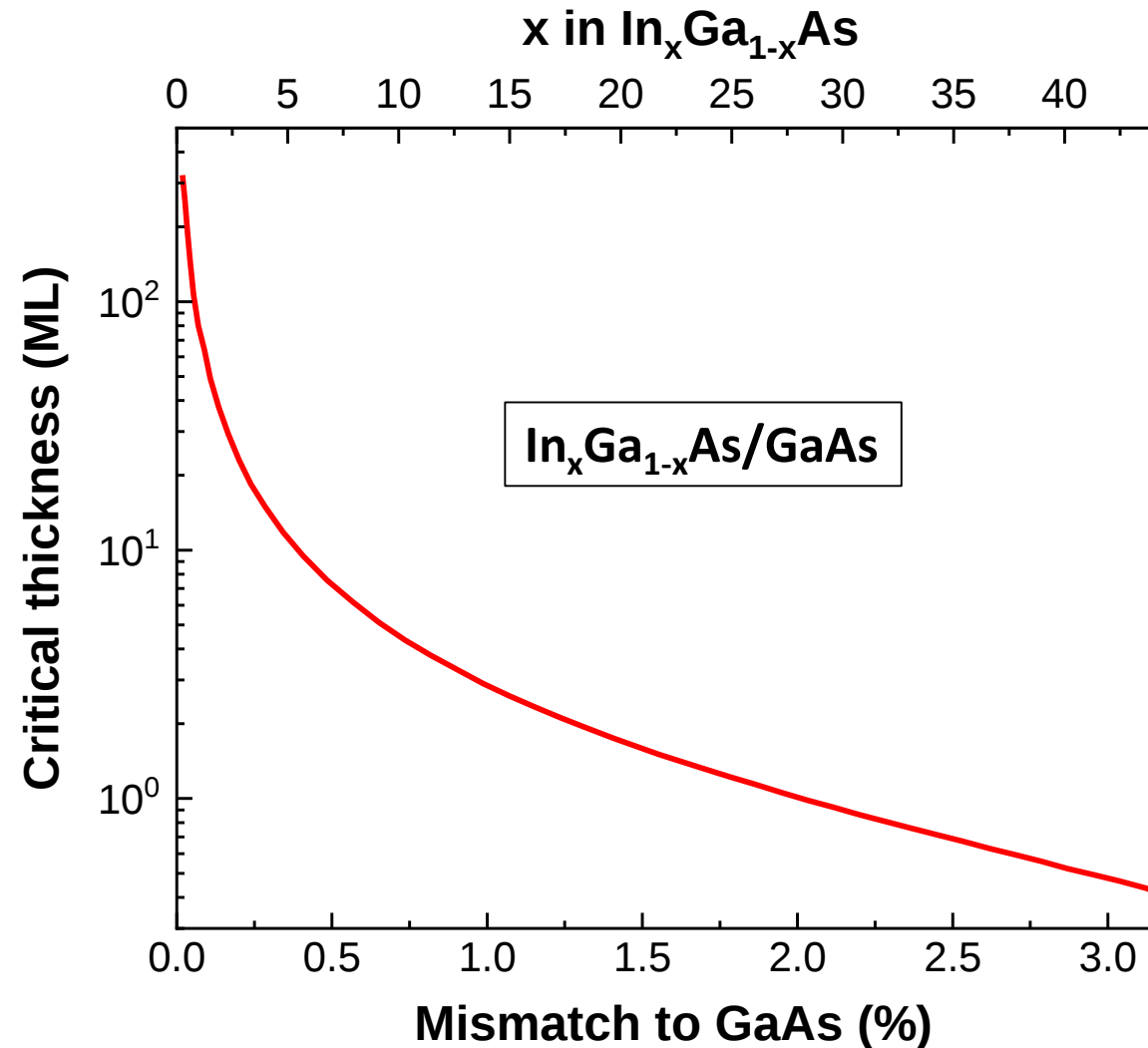
Dislocation mediated plastic relaxation



Strained 2D growth and plastic relaxation

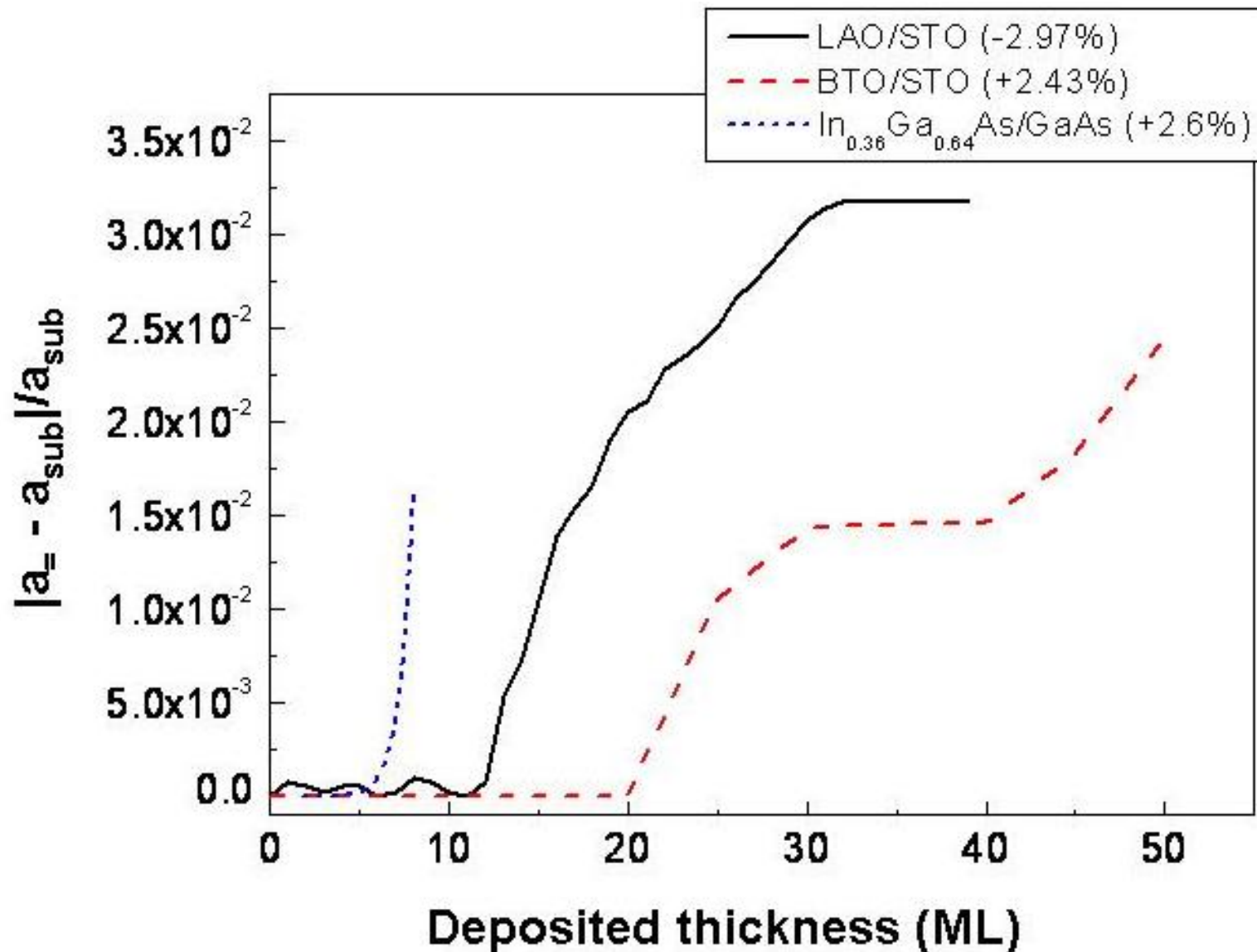
Dislocation mediated plastic relaxation

Critical thickness dependency to lattice mismatch



Strained 2D growth and plastic relaxation

Plastic relaxation : comparison of oxides and semiconductors



In_{0.36}Ga_{0.54}As/GaAs

$m = 2.6\%$

$Y \approx 85$

Polar covalent

BaTiO₃/SrTiO₃

$m = 2.4\%$

$Y \approx 67$

Ionic

LaAlO₃/SrTiO₃

$m = -2.9\%$

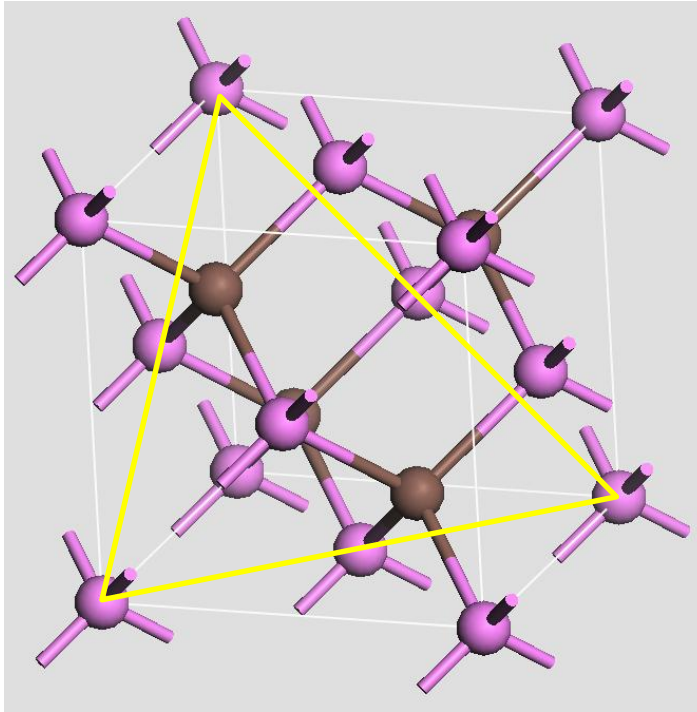
$Y \approx 331$

Ionic

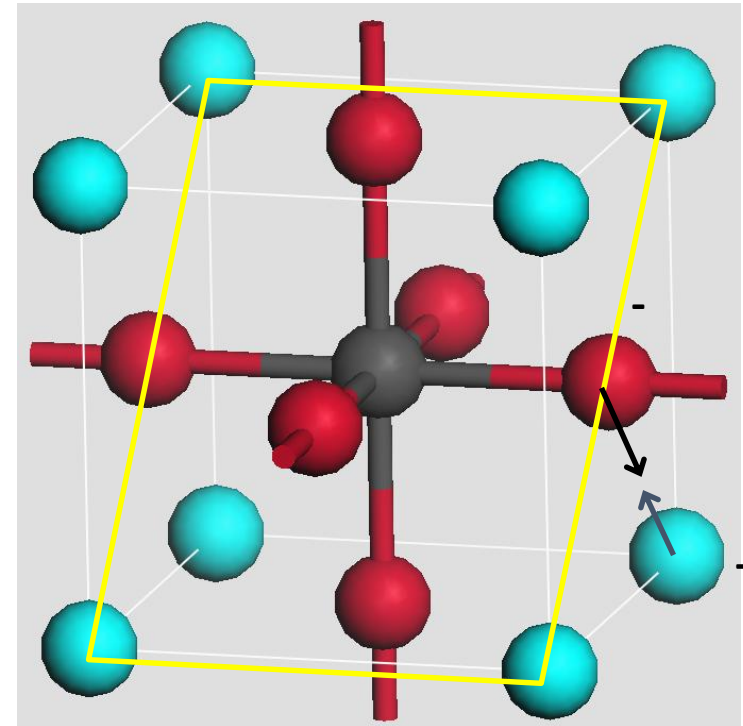
Strained 2D growth and plastic relaxation

Plastic relaxation : comparison of oxides and semiconductors

Zincblende
Gliding planes = $\{111\}$



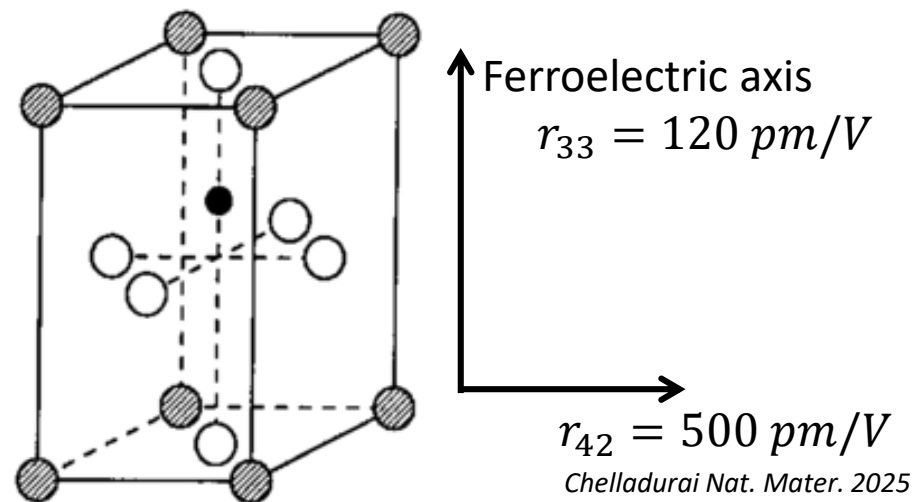
Perovskite
Gliding planes = $\{110\}$



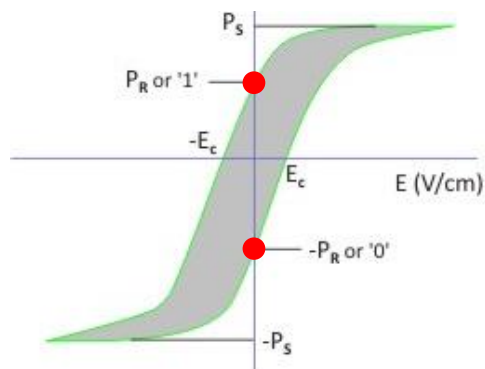
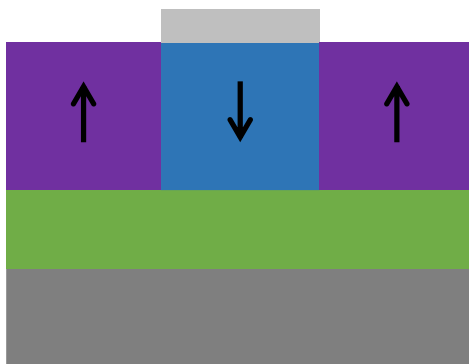
Partial charges hinder dislocation nucleation and propagation

Strained 2D growth and plastic relaxation

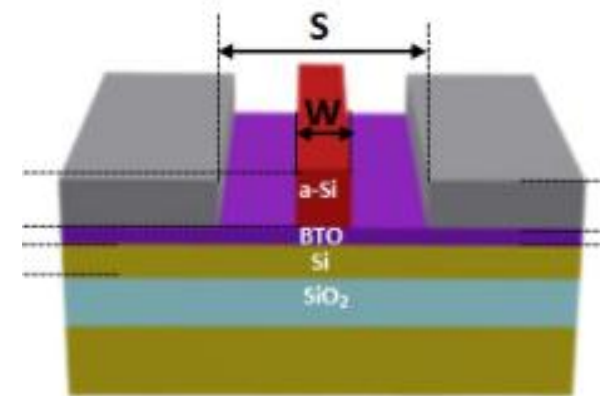
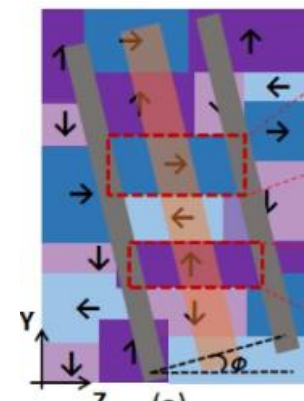
BaTiO₃ ferroelectric domain structure



Optimal domain structure for non volatile memory



Optimal domain structure for electro-optic modulator



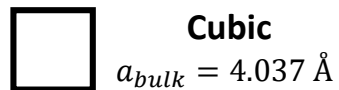
(LiNbO₃: $r_{33} = 15 \text{ pm/V}$, $r_{33} = 30 \text{ pm/V}$) 17

Strained 2D growth and plastic relaxation

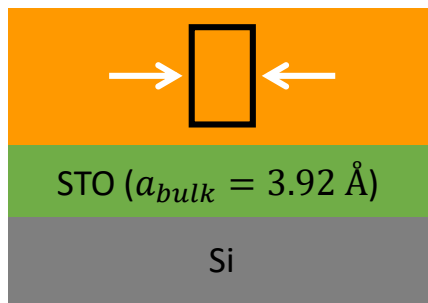
BaTiO₃ ferroelectric domain structure

@growth temperature (700°C)

Bulk BTO

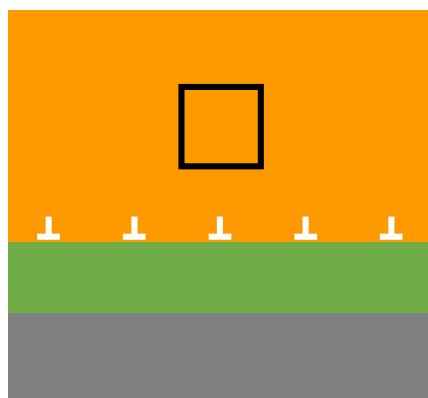


Compressive epi strain



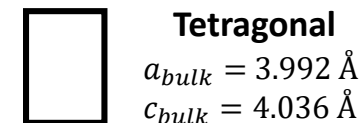
BTO/STO/Si
Thin ($< \sim 30 \text{ nm}$)

No epi strain

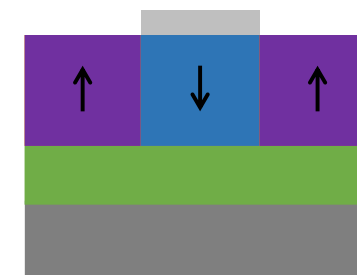
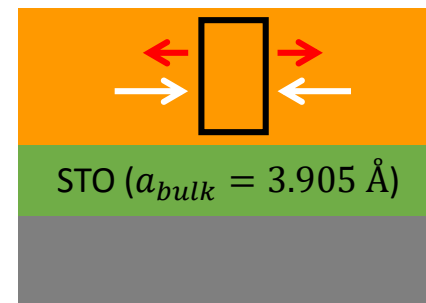


BTO/STO/Si
Thick ($> \sim 30 \text{ nm}$)

@room temperature



Compressive epi strain: optimal for non volatile memories



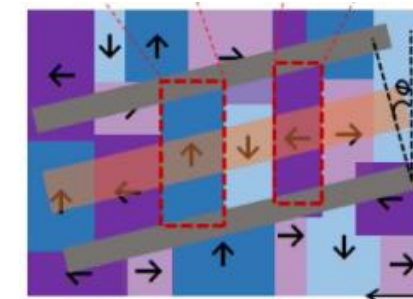
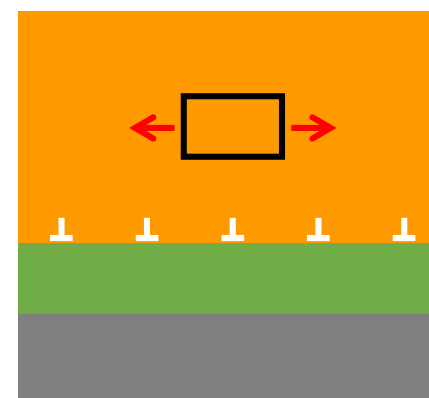
Cubic to tetragonal
phase transition

Post growth T
ramp

TEC

Si: $3 \times 10^{-6} \text{ K}^{-1}$
BTO: $1.3 \times 10^{-5} \text{ K}^{-1}$

Thermal tensile strain: optimal for electro-optic modulators

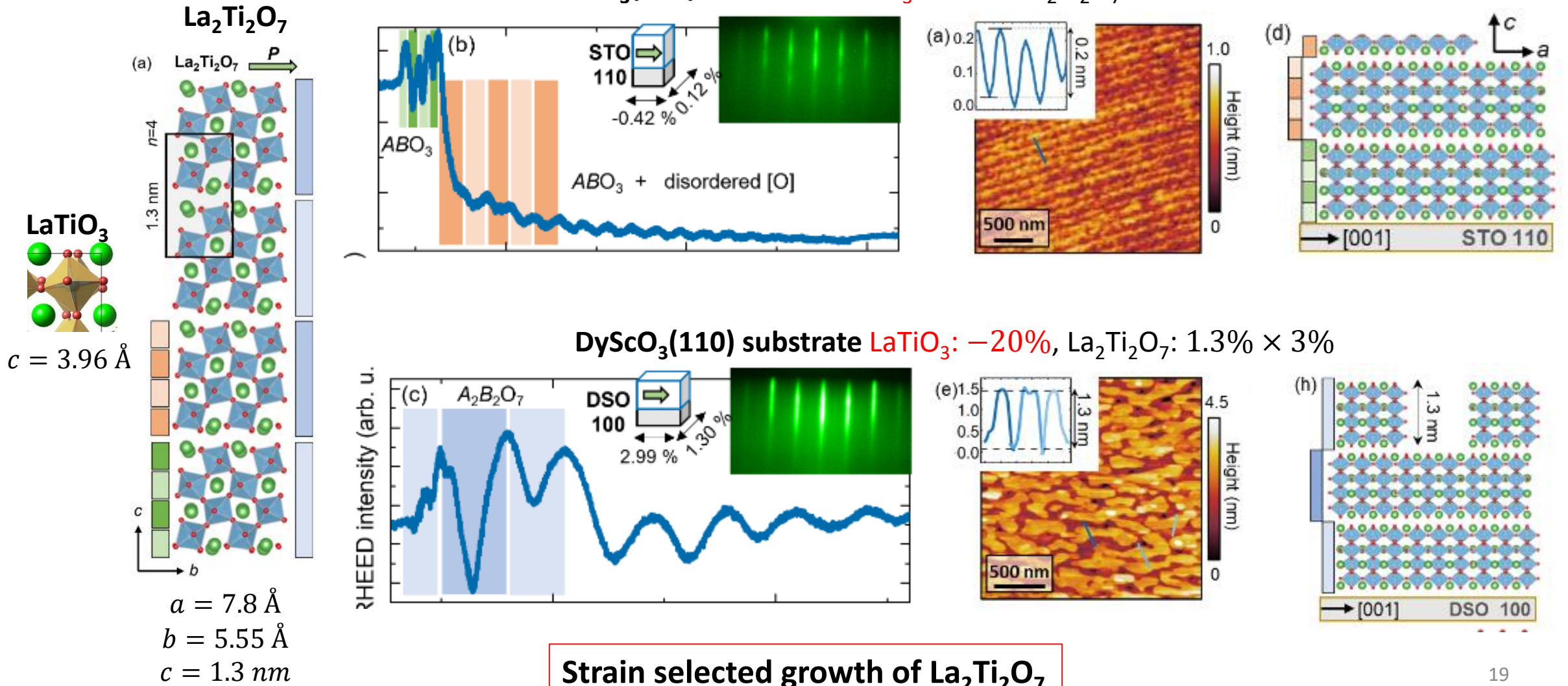


Strained 2D growth and plastic relaxation

Strain-assisted phase selection during Carpy-Gally compound growth

E. Gradauskaite et al., Adv. Mater. 37, 2416963 (2025)

SrTiO₃(110) substrate **LaTiO₃: 1.4 %**, La₂Ti₂O₇: 0.12% × −0.42%



Strain free 3D growth

- a. Strain-induced 3D growth: Stranski-Krastanov growth mode
- b. Interface induced 3D growth

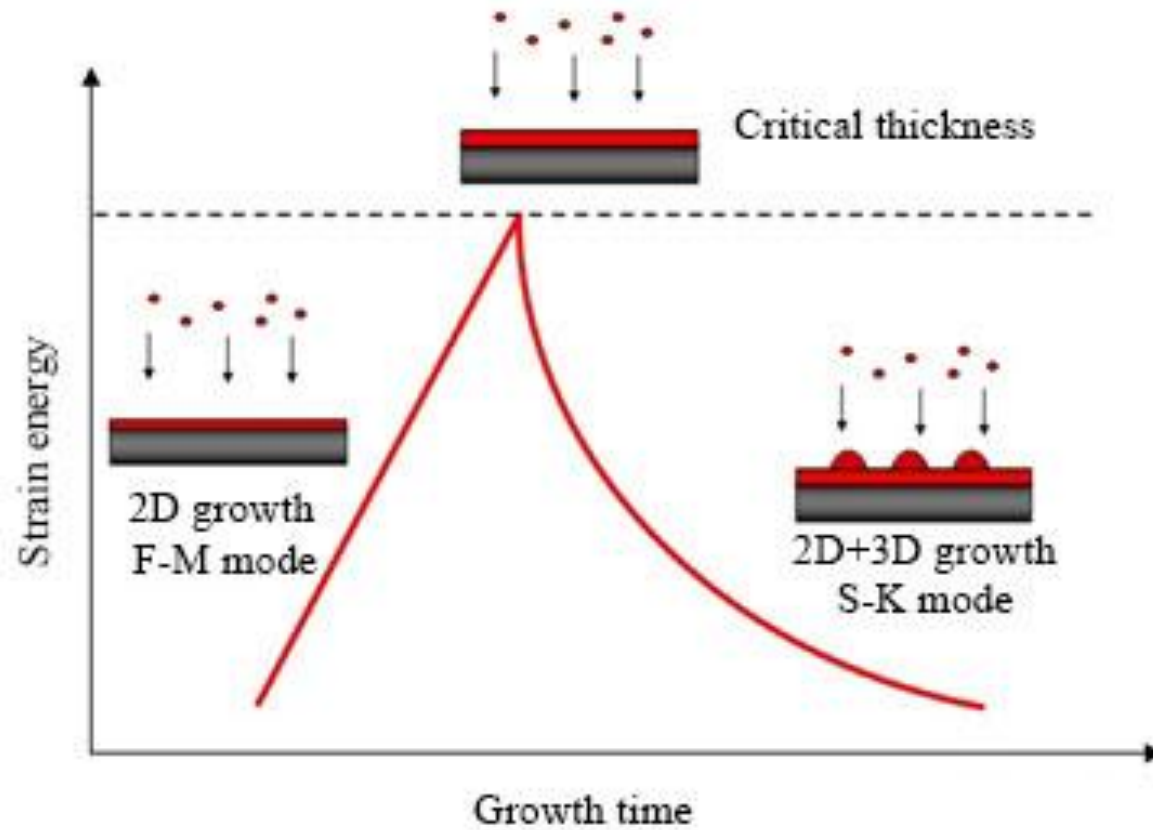
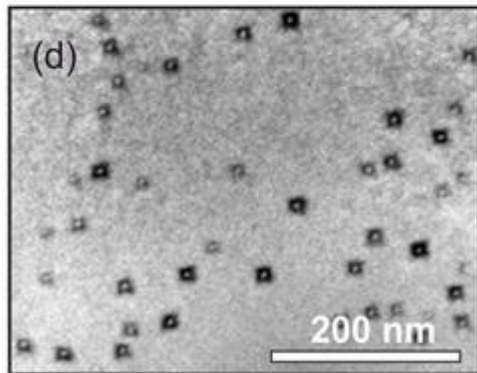
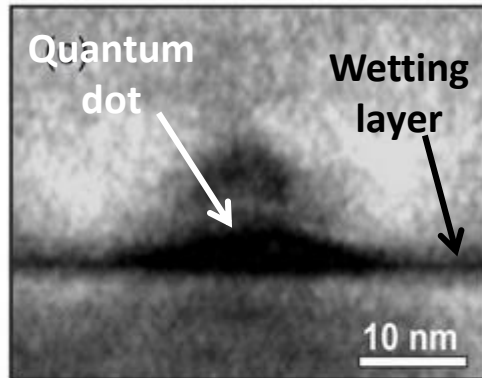
Strain free 3D growth

Strain induced 3D growth

Stranski-Krastanov growth mode

InAs/GaAs

Lattice mismatch $m = 7.16\%$
Strong adhesion, $\gamma_{AB} < 0$



Free surface strain relaxation

Strain free 3D growth

Interface induced 3D growth

GaP/Si

Lucci et al., Phys Rev Mat **2**, 060401(R), (2018)

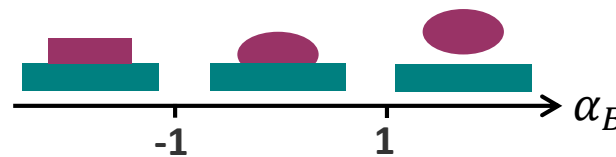
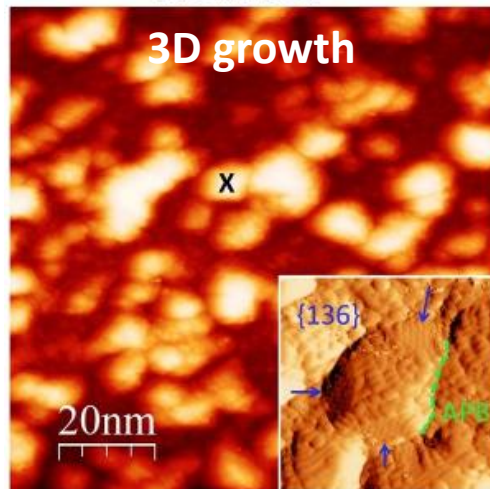
Low mismatch: $m = 0.37\%$

$$\gamma_{Si} = 1.4 \text{ J/m}^2$$

$$\gamma_{GaP} = 0.92 \text{ J/m}^2$$

$$\gamma_{GaP/Si} > 0.5 \text{ J/m}^2 \text{ (DFT)}$$

$$\alpha_B = \frac{\gamma_{GaP/Si} - \gamma_{Si}}{\gamma_{GaP}} > -0.98 > -1$$



Ge/BaTiO₃

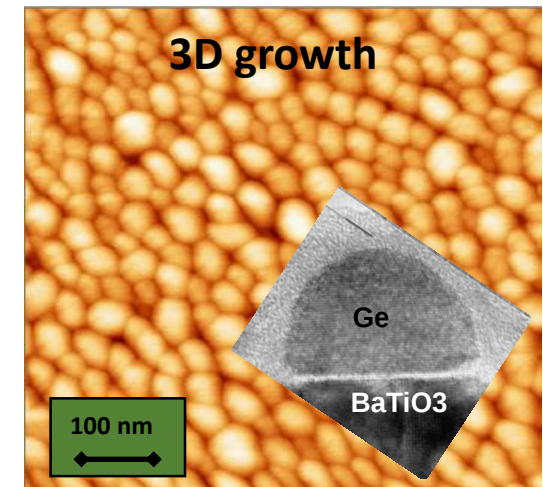
Low mismatch: $m = 0.2\%$

$$\gamma_{BTO} = 1 \text{ J/m}^2$$

$$\gamma_{Ge} = 0.8 \text{ J/m}^2$$

Instable Ge-O interface bonds

$$\alpha_B = \frac{\gamma_{Ge/BTO} - \gamma_{BTO}}{\gamma_{Ge}} > -1$$



Highly dissimilar epitaxial systems

a. Indirect epitaxial relationships

- Driven by lattice mismatch
- Driven by interface energy

b. Mismatch accommodation via interfacial dislocation networks

- Dislocation entry at the early stages of the growth
- Defects formed during coalescence

b. Interface chemical reactions

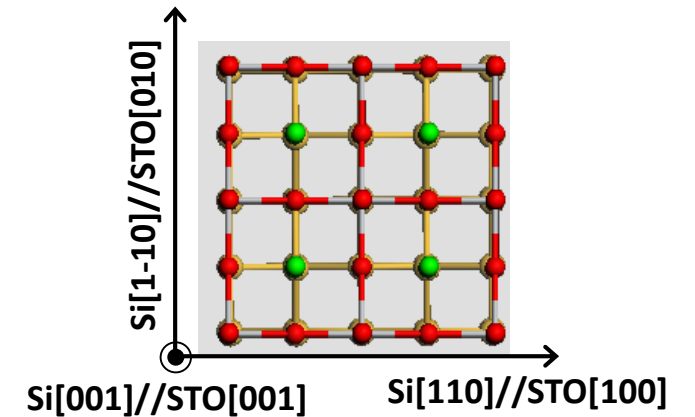
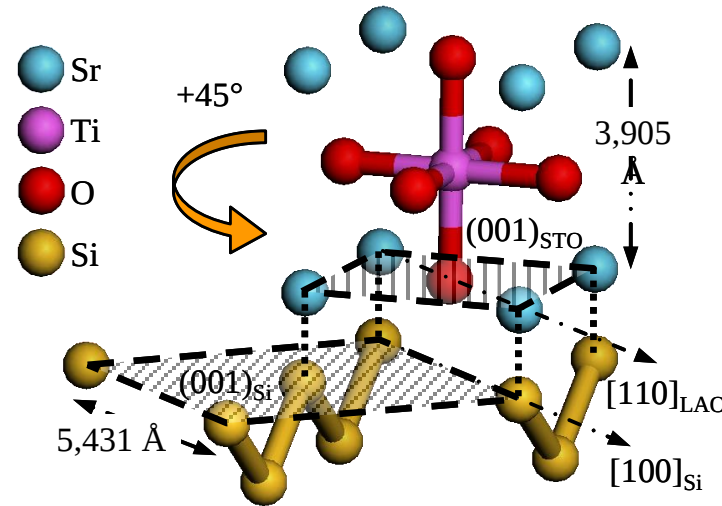
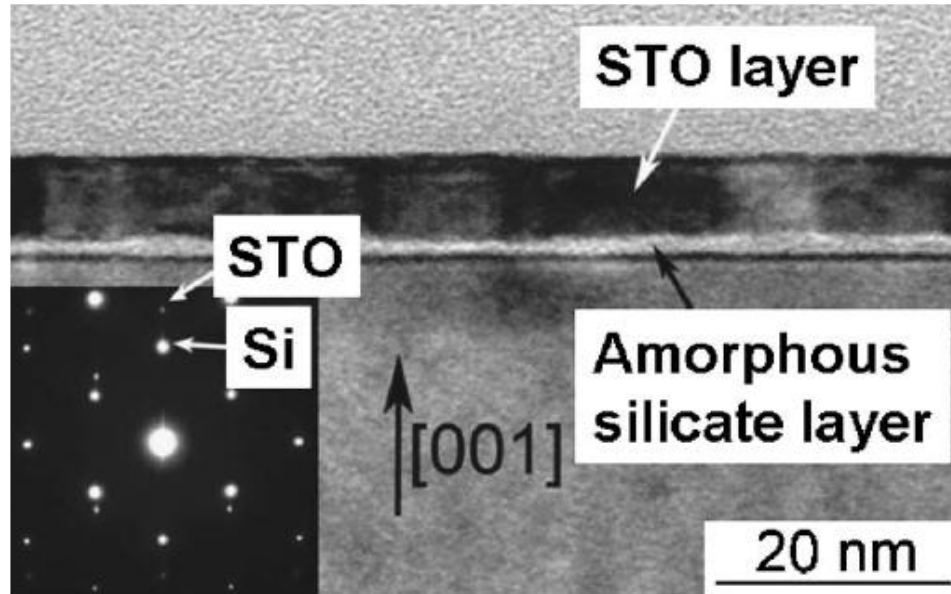
- Growth window
- SrTiO_3/Si : the knitting machine

Highly dissimilar epitaxial systems

Indirect epitaxial relationships

$\text{SrTiO}_3/\text{Si}(001)$: epitaxial relationship driven by lattice mismatch

R. McKee et al., Phys. Rev. Lett. 81, 3014 (1998)

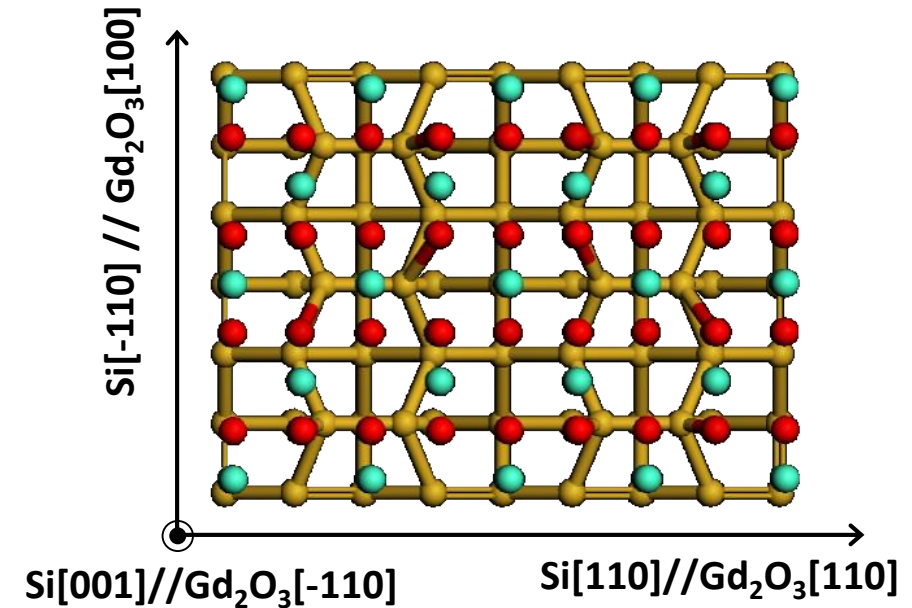
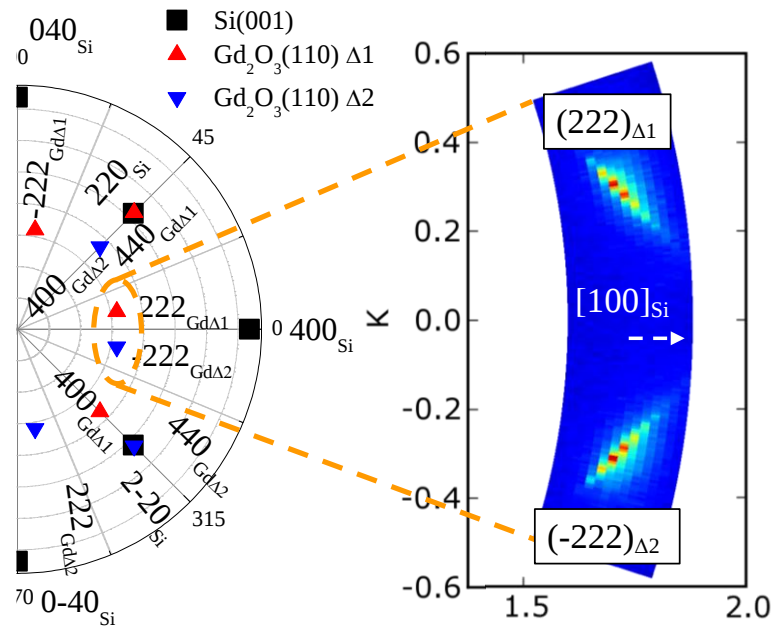


« Cube on cube »: $m = 48\%$
 45° rotation: $m = 1.7\%$

Highly dissimilar epitaxial systems

Indirect epitaxial relationships

$\text{Gd}_2\text{O}_3/\text{Si}(001)$: epitaxial relationship driven by interface energy



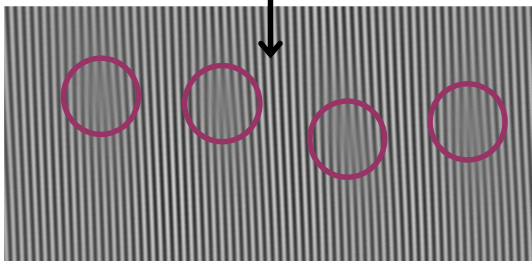
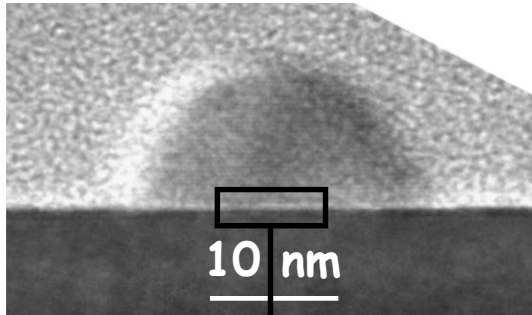
« Cube on cube »: $m = -0.44\%$, **62% dangling bonds**
 Indirect: $m = -0.44 \times 5.6\%$, **25% dangling bonds**

Highly dissimilar epitaxial systems

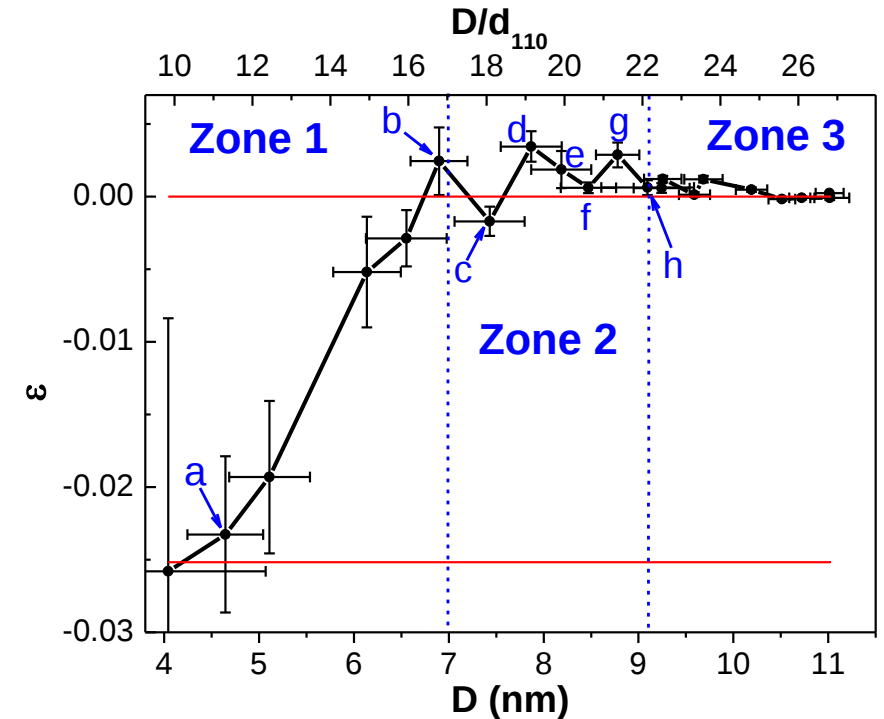
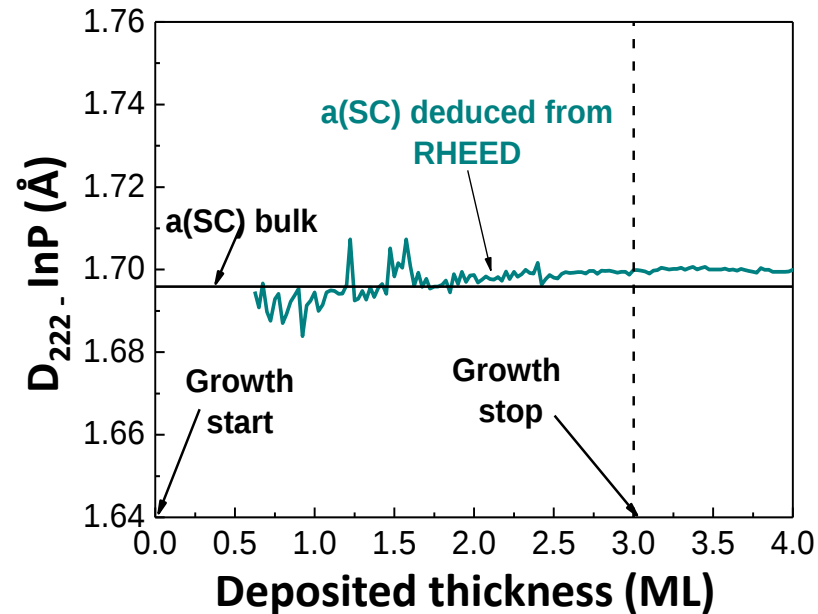
Mismatch accommodation via interfacial dislocation networks

InP/SrTiO₃ and Ge/SrTiO₃

Danescu et al., Appl. Phys. Lett. **103**, 021602 (2013), Saint-Girons et al., Appl. Phys. Lett. **92**, 241907 (2008)



Indirect epitaxial relationship
 $\langle 110 \rangle \text{InP}(001) // \langle 100 \rangle \text{STO}(001)$
 $D/a = 6.3\%$



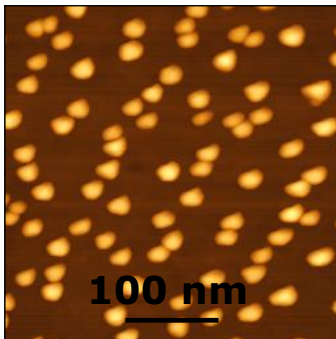
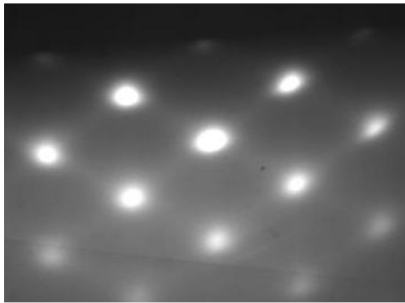
Full mismatch accommodation by interface dislocations formed at the very early stages of the growth

Highly dissimilar epitaxial systems

Mismatch accommodation via interfacial dislocation networks

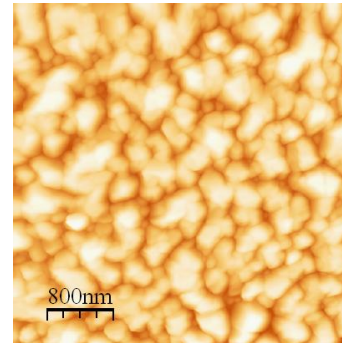
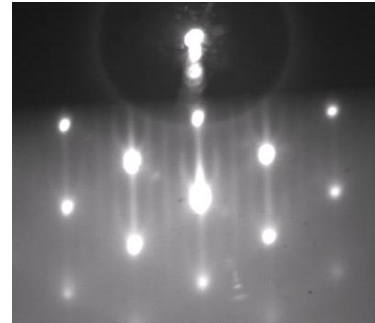
InP/SrTiO₃: coalescence

Step 1
islands



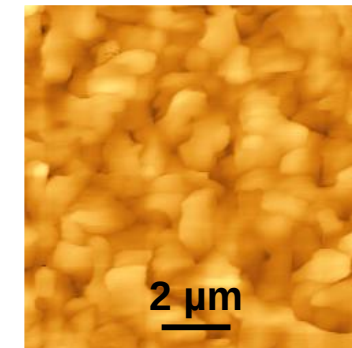
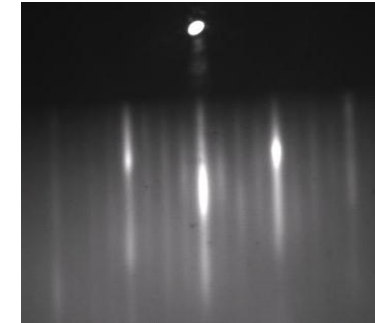
Low temperature (400°C)
High P (10⁻⁵ Torr)
→ condensation

Step 2
coalescence



High temperature (510°C)
Low P (10⁻⁶ Torr)
→ Surface diffusion

Step 3
growth



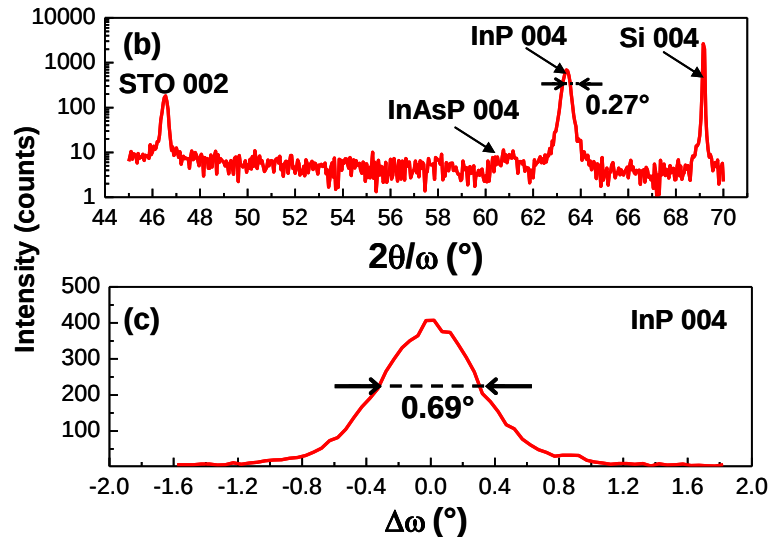
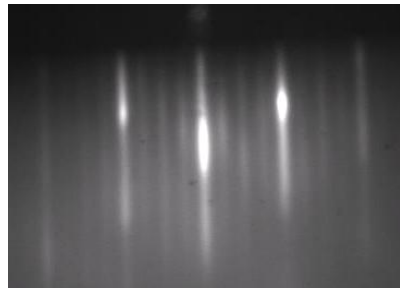
480°C – 10⁻⁵ Torr
→ Standard InP conditions

Highly dissimilar epitaxial systems

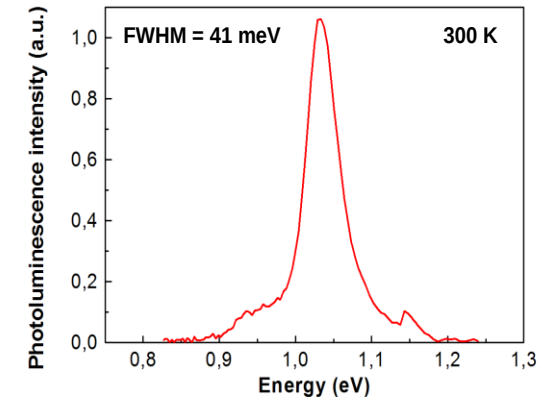
Mismatch accommodation via interfacial dislocation networks

Integration of InP based heterostructures on Si using SrTiO₃ templates

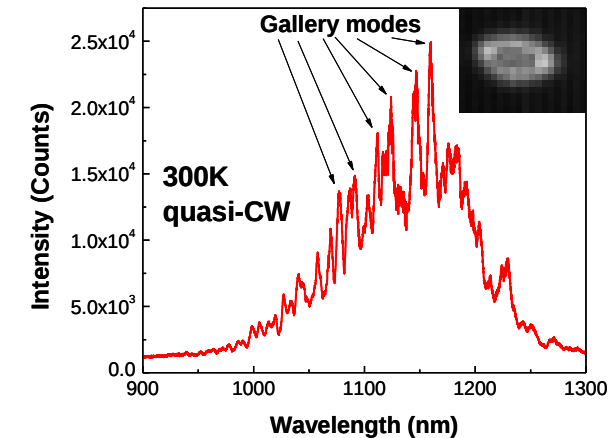
500 nm thick InAsP/InP/STO/Si quantum well heterostructure



Gobaut et al, Appl. Phys. Lett. **97**, 201908, (2010)



Intense and narrow PL signal @300K



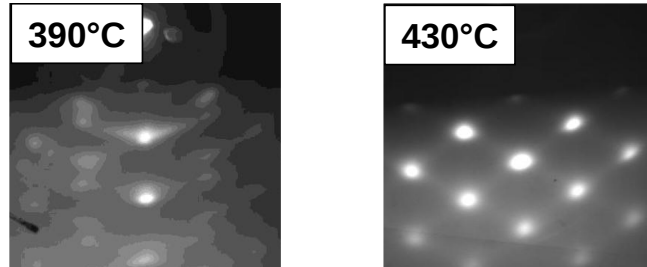
Microdisk laser : light amplification but no lasing

Highly dissimilar epitaxial systems

Mismatch accommodation via interfacial dislocation networks

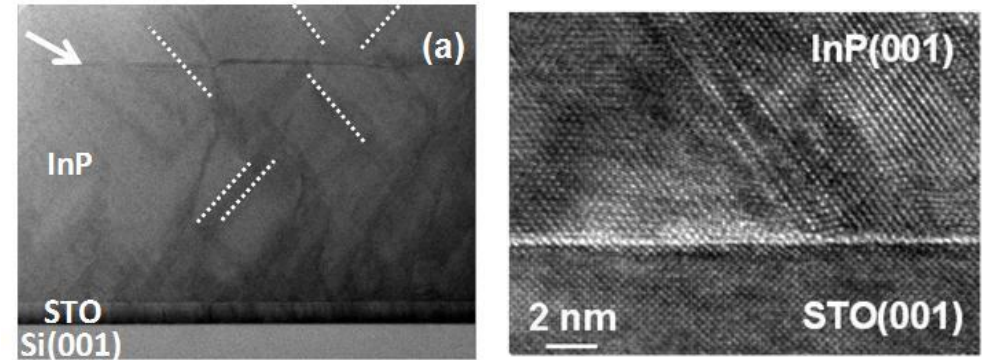
Integration of InP based heterostructures on Si using SrTiO₃ templates: defects and limitations

InP orientation depends on growth conditions



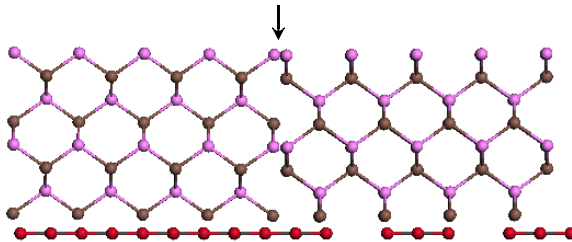
InP(111) / STO(001) InP(001) / STO(001)

Twins are formed due to strong InP dewetting

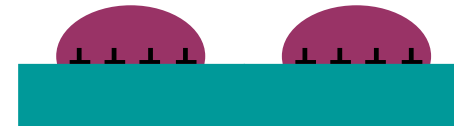


Antiphase Domains

Domain boundary



Commensurability and interface dislocation network



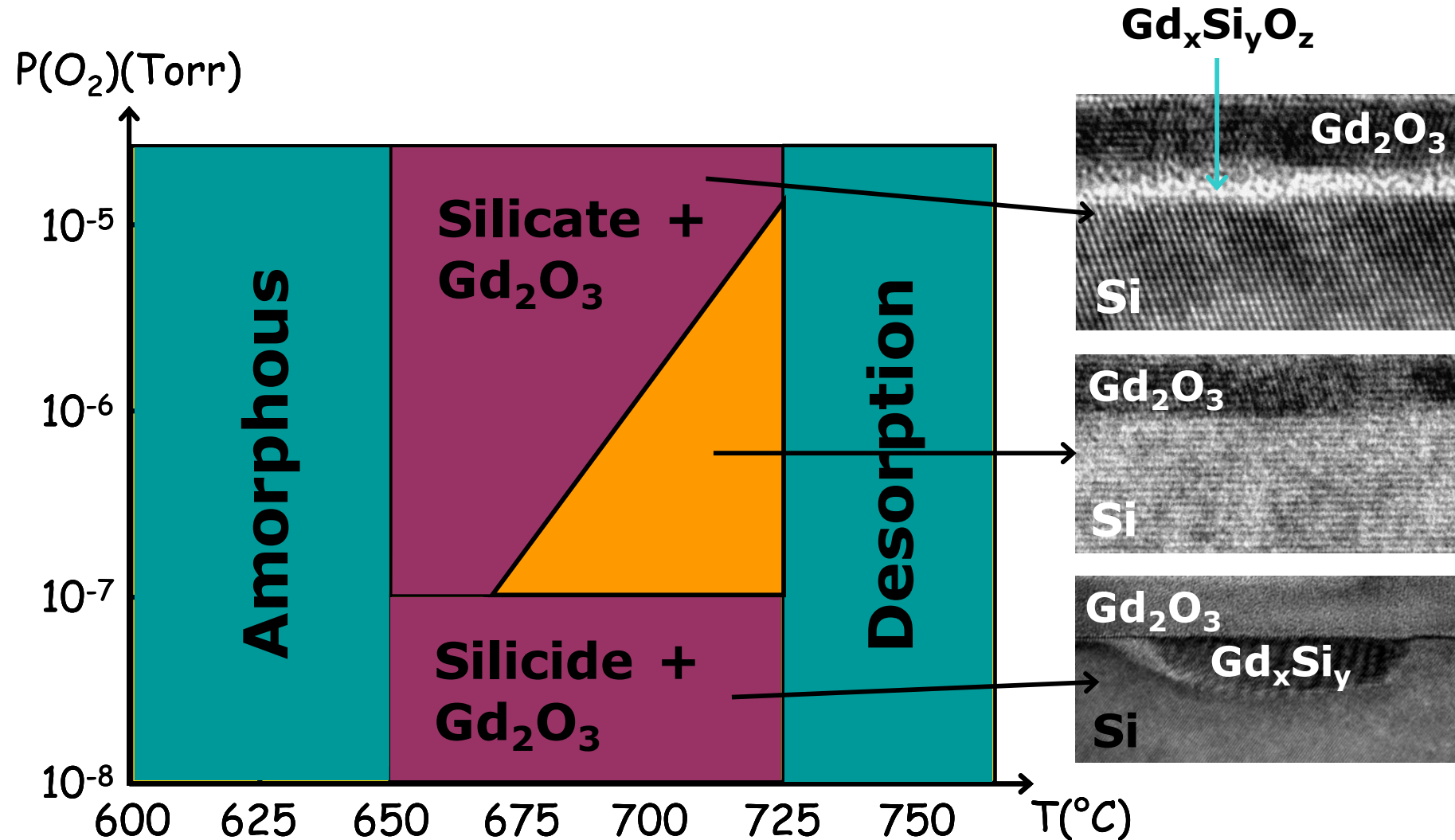
There is no reason why lateral registry between neighbouring islands should be maintained

➔ Defect formation during coalescence

Highly dissimilar epitaxial systems

Interface chemical reactions

Growth window: $\text{Gd}_2\text{O}_3/\text{Si}(111)$



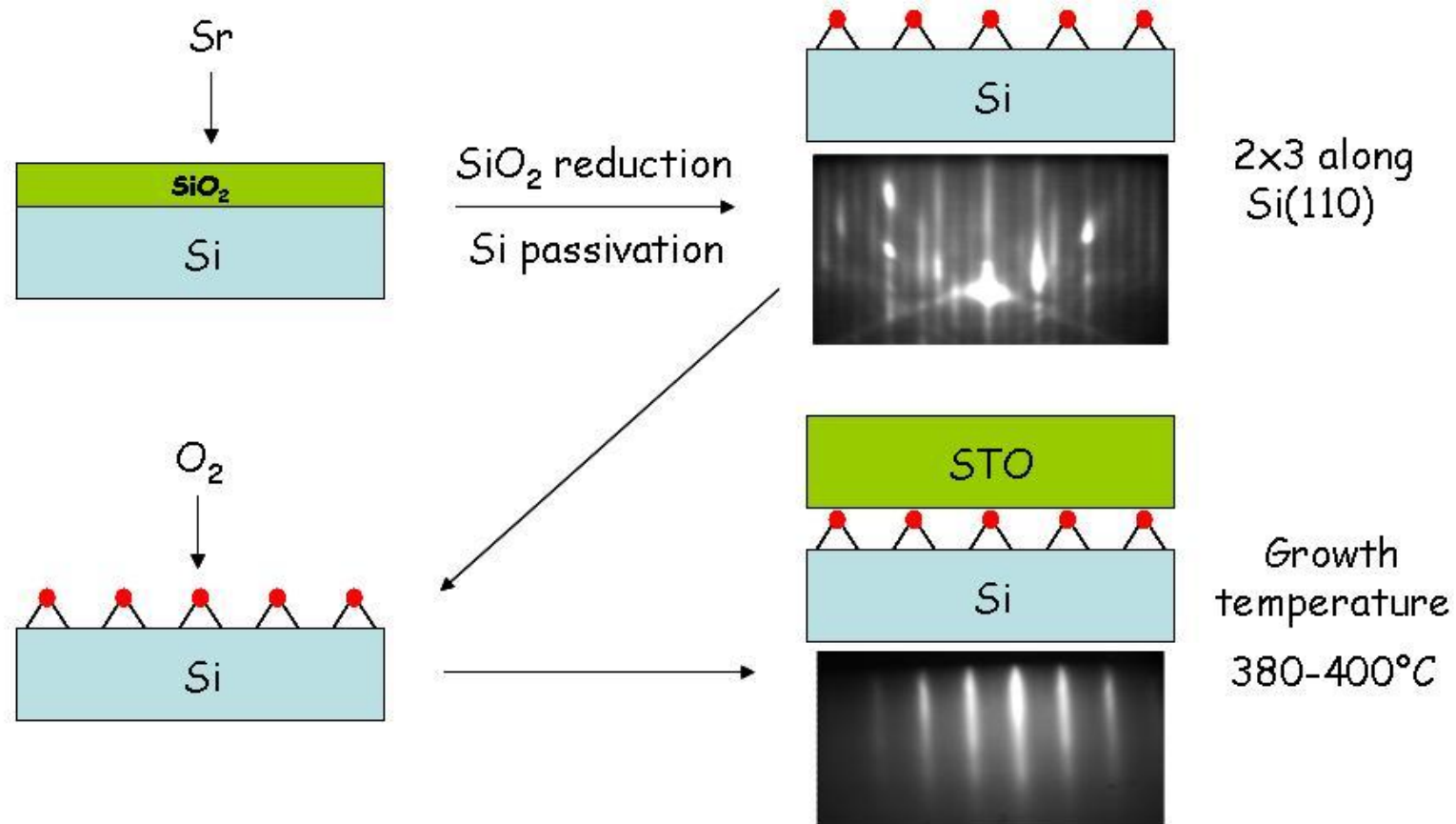
Highly dissimilar epitaxial systems

Interface chemical reactions

Growth window: $\text{SrTiO}_3/\text{Si}(001)$

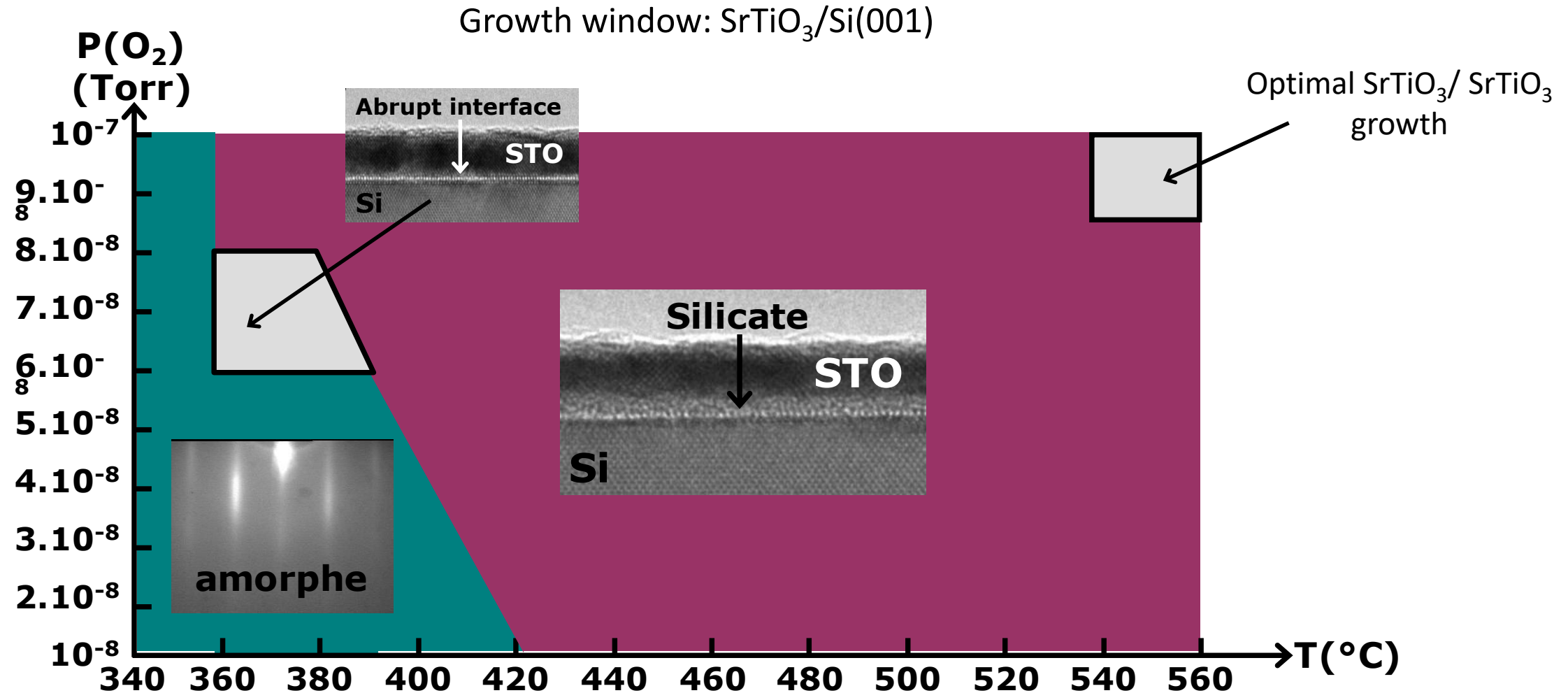
The direct growth of SrTiO_3 on Si is impossible : silicates/silicides formation

→ Si surface passivation



Highly dissimilar epitaxial systems

Interface chemical reactions



Narrow growth window

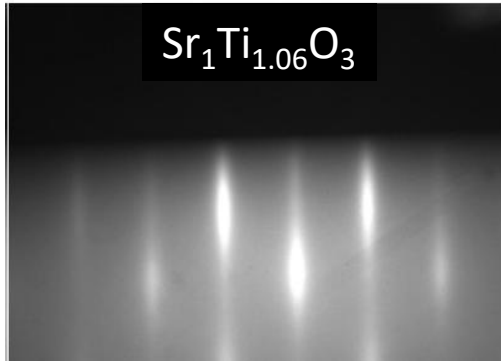
Highly dissimilar epitaxial systems

Interface chemical reactions

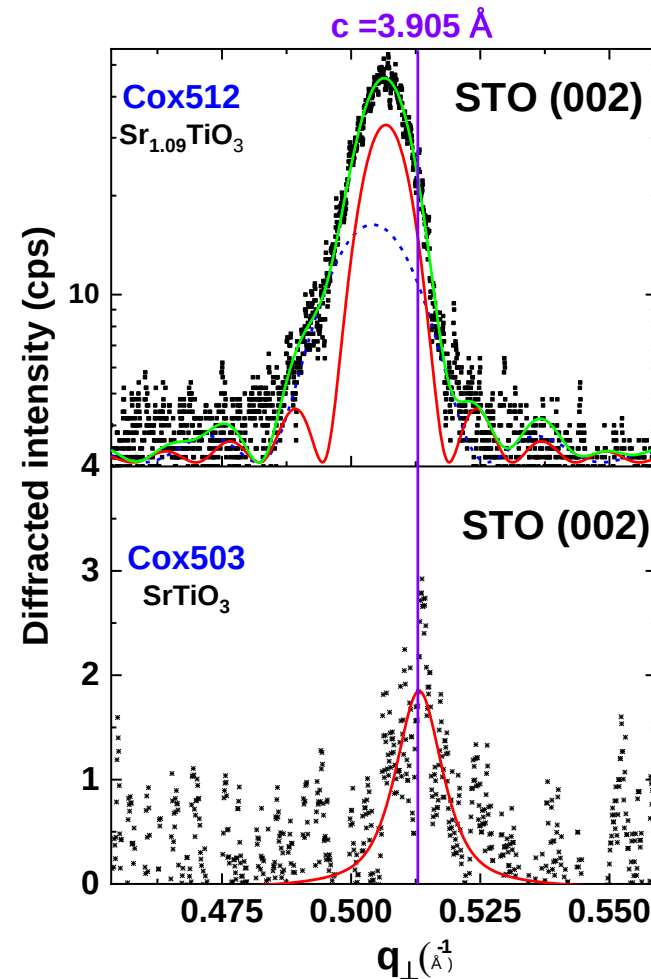
$\text{SrTiO}_3/\text{Si}(001)$: crystallization process

Saint-Girons et al., Chem. Mat. 28, 5347 (2016)

RHEED



XRD



SrTiO_3 growth process :

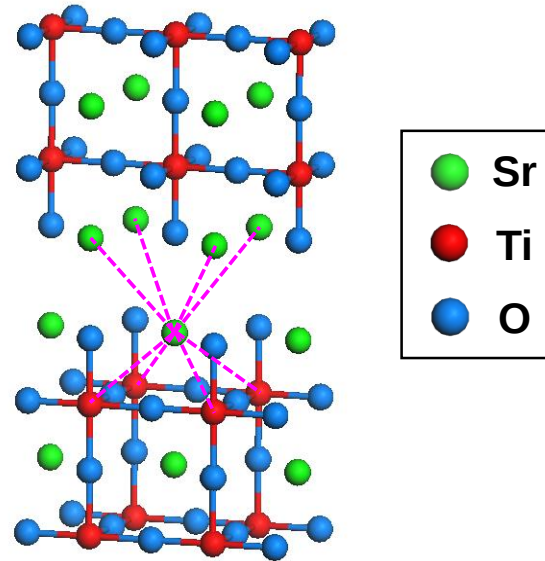
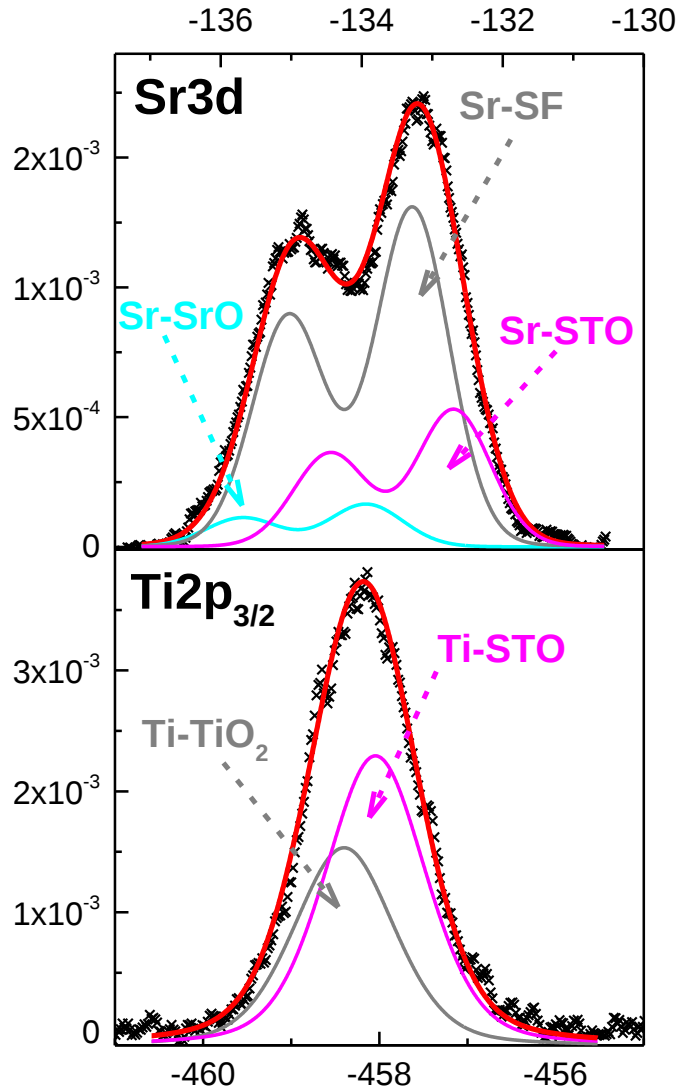
- 1-deposition of (partially) amorphous STO @300°C
- 2-crystallization by annealing @450°C under UHV

SrTiO_3 crystallization
requires an excess of Sr

Highly dissimilar epitaxial systems

Interface chemical reactions

$\text{SrTiO}_3/\text{Si}(001)$: crystallization process



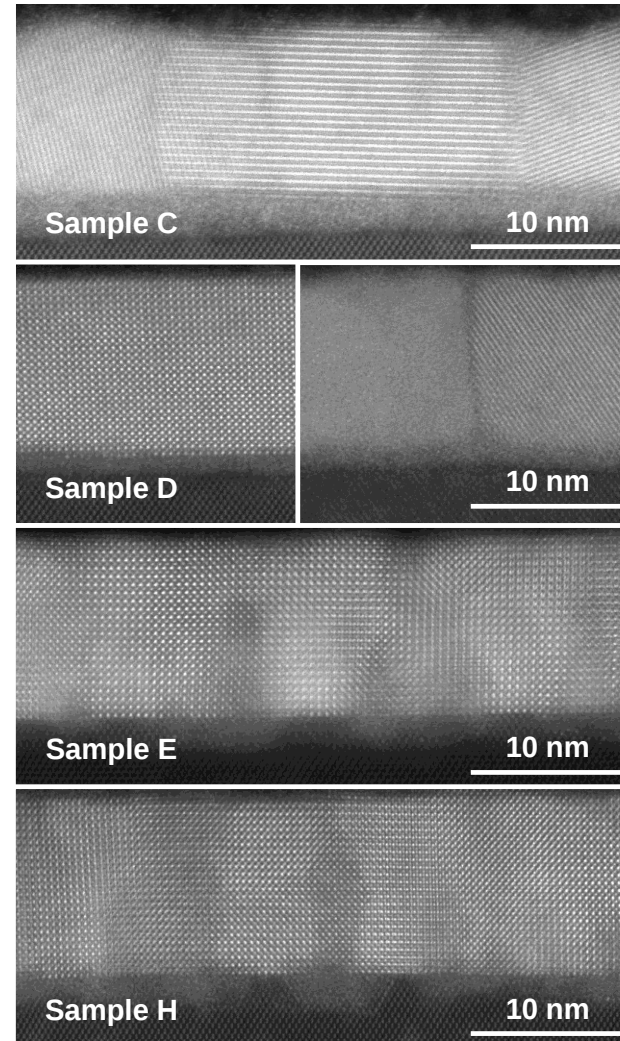
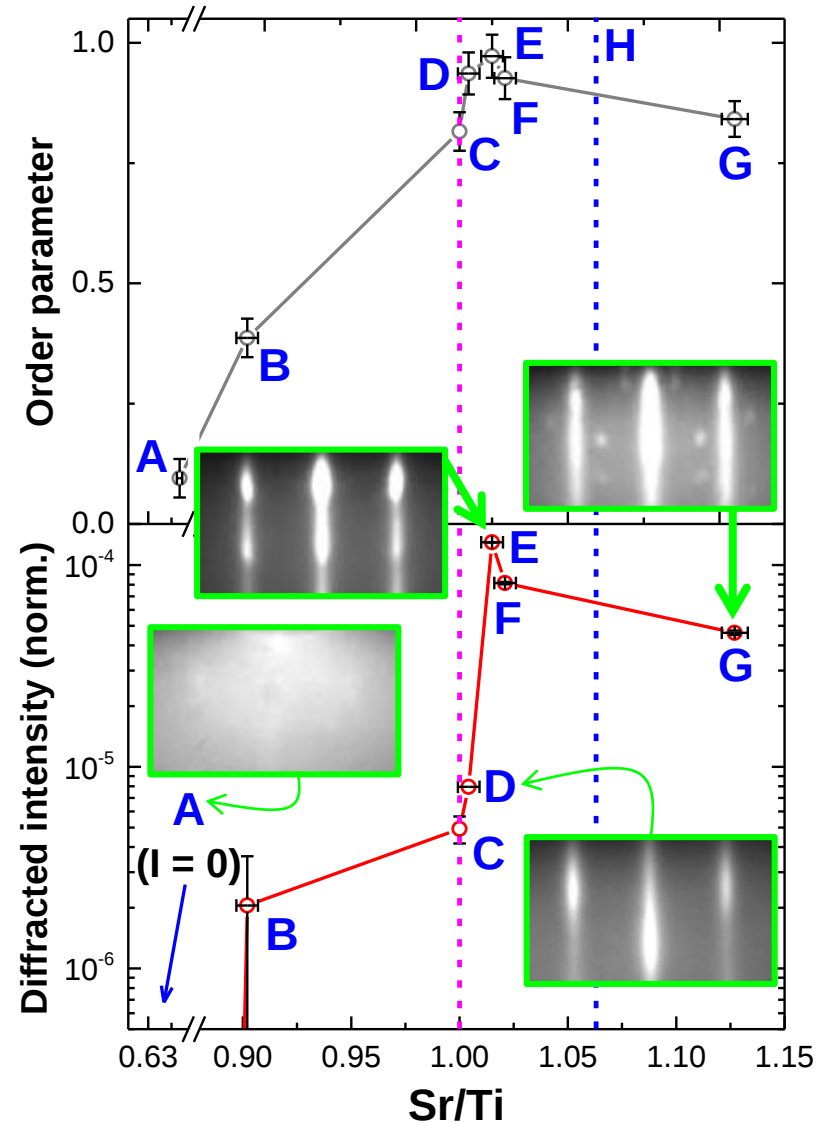
Order parameter P ($P=1$ for pure STO)

$$P = \frac{n(\text{Sr} - \text{Ti}) - (n(\text{Sr} - \text{Sr}) + n(\text{Ti} - \text{Ti}))}{n(\text{Sr} - \text{Ti}) + n(\text{Sr} - \text{Sr}) + n(\text{Ti} - \text{Ti})}$$

« STO » is composed of
STO, SF, SrO and TiO₂ ~ **partial demixion**

Highly dissimilar epitaxial systems

Interface chemical reactions

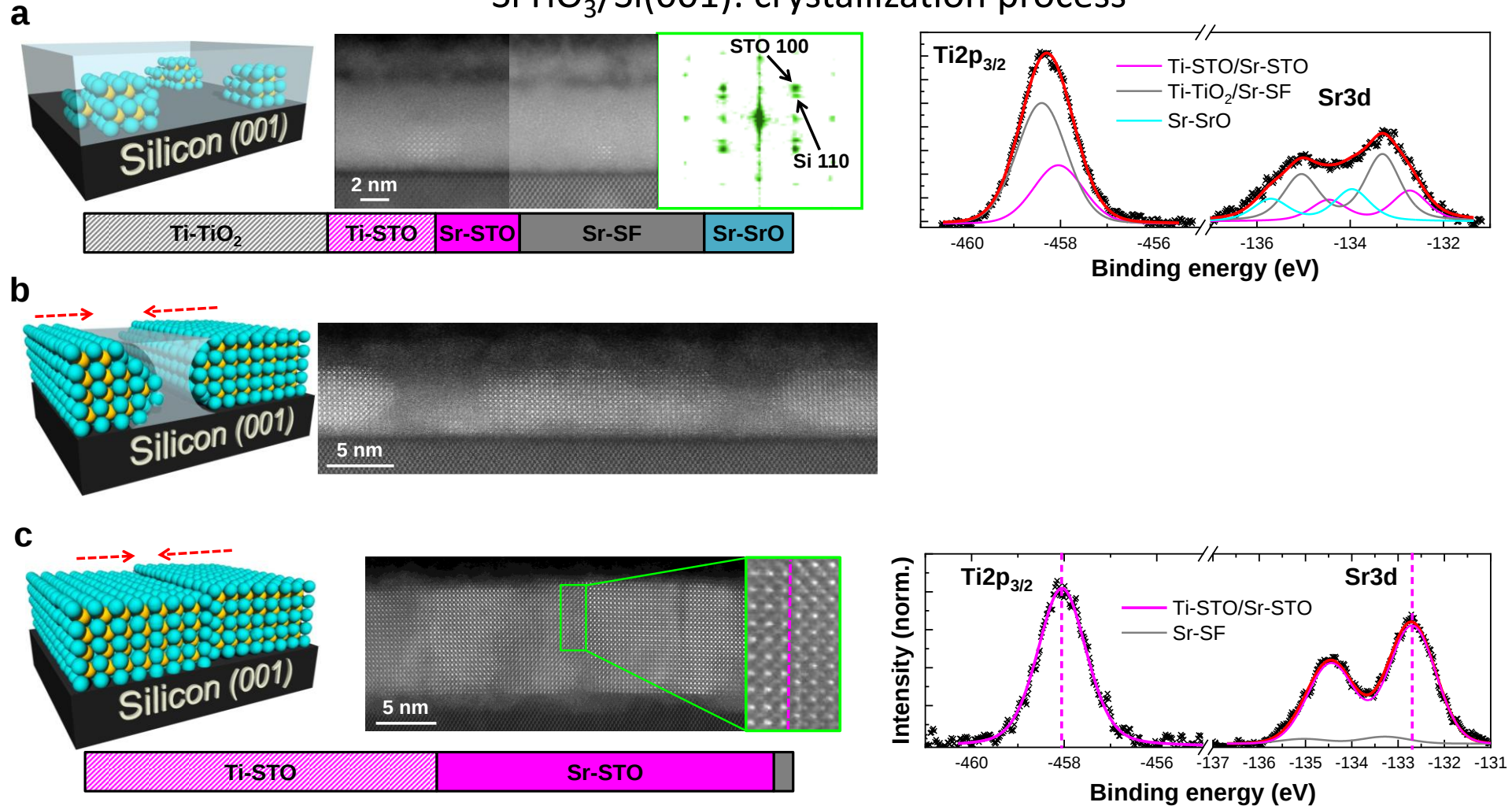


Sr/Ti = 1.015:
optimal
chemical and
structural order

Highly dissimilar epitaxial systems

Interface chemical reactions

$\text{SrTiO}_3/\text{Si}(001)$: crystallization process

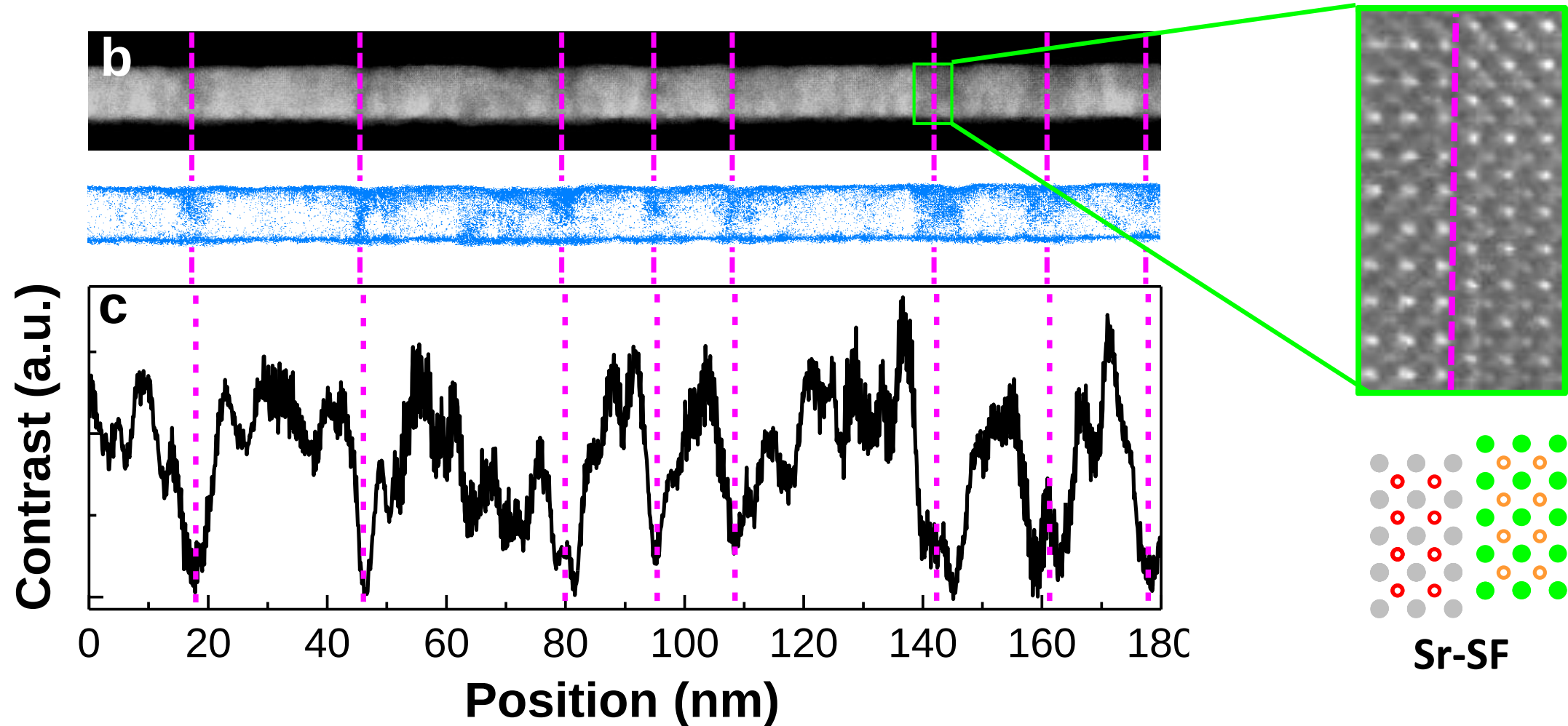


- Crystallization = propagation of a Sr-rich front
- SF are formed when neighboring crystallites coalesce

Highly dissimilar epitaxial systems

Interface chemical reactions

$\text{SrTiO}_3/\text{Si}(001)$: crystallization process



After crystallization: antiphase domain morphology

Conclusion and future challenges

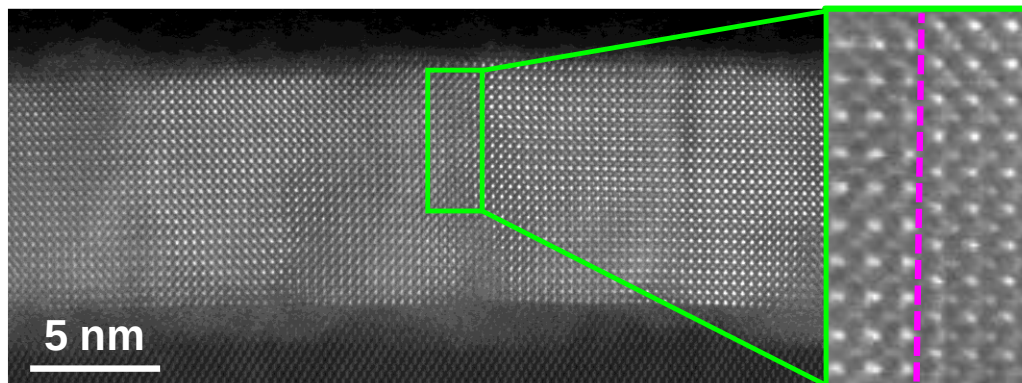
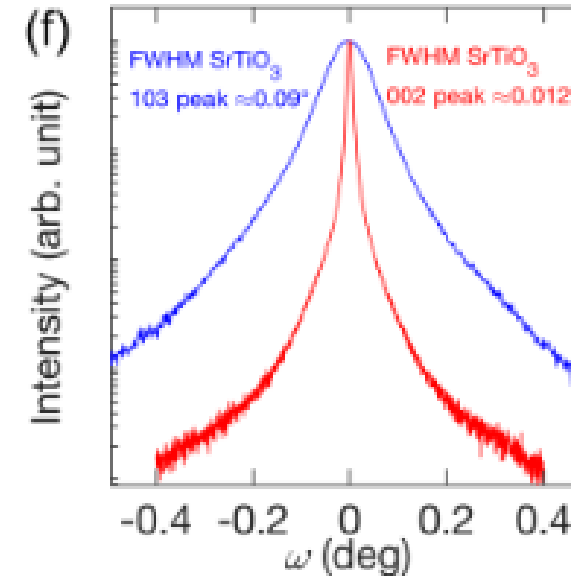
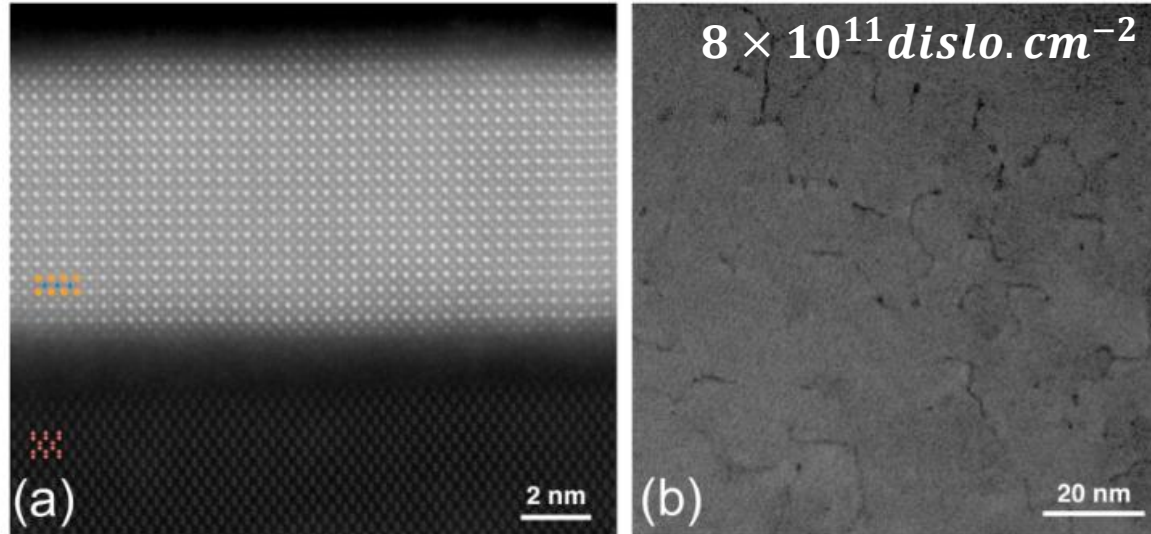
- a. Oxide/Si templates: a mature integration technology
- b. MBE for oxide growth: advantages and challenges

Conclusion and future challenges

Oxide/Si templates: a mature integration technology

SrTiO_3/Si : good structural quality can be achieved

Wang *Phys. Rev. Mat.* **3**, 073403 (2019)



Dislocations and out-of-plane antiphase boundaries resulting from the crystallization process limit the structural quality

Conclusion and future challenges

Oxide/Si templates: a mature integration technology

A short review

- **TiO₂** → photocatalytic water splitting
Choi J. Appl. Phys. 111, 064112, (2012)
- **LaCoO₃** → thermoelectric, ferromagnetic
Posadas Appl. Phys. Lett. 98, 053104, (2011)
- **BaTiO₃** → nanoelectronic/memories/photonics
Niu Microelec. Eng. 88, 1232, (2011)
Dubourdieu Nature Nanotech. 8, 881, (2013)
Scigaj Appl. Phys. Lett. 109, 122903 (2016)
Eltes J. Ligthw. Tec. 37, 1456 (2019)
- **LaAlO₃** → high-k dielectric
Mi Appl. Phys. Lett. 90, 181925, (2007)
- **BiFeO₃** → multiferroic
Wang Appl. Phys. Lett. 85, 2574, (2004)
- **PZT** → RF-filters/MEMS
Lin Appl. Phys. Lett. 78, 2034, (2001)
- **LaSrMnO₃** → spintronics, sensors
Le Bourdais J. Appl. Phys. 118, 124509, (2015)
- **PMN-PT** → RF-filters/MEMS
Baek Science 334, 6058 (2011)
- **Diamond** → high power electronics
Arnault Diamond Rel. Mat. 105, 107768 (2020)
- **HfZrO₂** → nanoelectronics
Song Nanoscale 15, 222901 (2023)
- **LiNbO₃** → photonics
Bartasyte, Nanotechnology 35, (2024)
- **Li₄Ti₅O₁₂** → batteries
Lacey, ACS Applied Mat. Int. 15, 1535 (2022)
- ...



Commercial BTO/STO/Si
electro-optic modulator

Conclusion and future challenges

MBE for oxide growth: advantages and challenges

MBE presents unique features for oxide growth

- Flexible composition control
- Interface engineering → oxide/semiconductor integration
- Heterostructures and superlattices
- High structural quality

Reproducible composition control under oxygen remains challenging

- Source drift due to elemental load oxidation
- Flux measurement under oxygen is challenging

Developing strategies for real time flux measurement and control under oxygen is the key to further improve oxide MBE process reliability